

Structural Integrity and Reliability
of Advanced Materials obtained
through additive Manufacturing

SIRAMMI

H2020-WIDESPREAD-2018



**European
Commission**

1st Winter School on
**Trends on Additive Manufacturing for
Engineering Applications**
24-28 January 2021

Review on AM of polymeric materials

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OUTLINE

1. INTRODUCTION & OBJECTIVES
2. BASIC CONCEPTS ON THE MECHANICS OF POLYMERS
3. AM TECHNOLOGIES FOR POLYMERS
4. AM PROCESS PARAMETERS IN POLYMERS
5. SELECTIVE LASER SINTERING (SLS): EXP/SIMULATIONS
6. PHOTOPOLYMERIZATION (SLA, DLP): EXP/SIMULATIONS
7. CONCLUSIONS

1. INTRODUCTION & OBJECTIVES

- Polymers are among the most diffused synthetic materials in modern applications.
- Their good mechanical properties, lightweight and low cost have enhanced their diffusion among the most diverse application fields.
- Their underlying microstructure provides a mechanical behavior mimicking that of biological materials (tissues, organs, etc.)
- They can be easily printed by using different technologies to obtain complex objects even at very small dimensional scale

- Understand the **microstructure** of a polymer
- Presents some basic concepts of the **mechanics** of polymers
- Illustrate the **main AM technologies used for polymers**
- Presents some in-depth aspects related to **SLS (*selective laser sintering*) technology**
- Presents some in-depth aspects related to **SLA (*stereolithography*) technology**

2. BASIC CONCEPTS ON THE MECHANICS OF POLYMERS

SOME CONCEPTS ON THE PHYSICS & MECHANICS OF POLYMERS

- **POLYMER MICROSTRUCTURE**
- **CHAINS AND CROSS-LINKS**
- **DEFORMATION OF POLYMERS**
- **DEFORMATION ENERGY (ENTROPIC)**

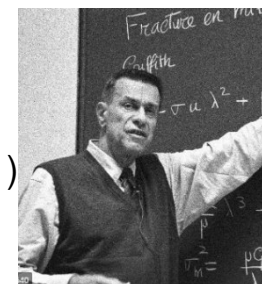
Paul Flory

Nobel prize in Chem. (1974)
For physical chemistry of
macromolecules



P.G. De Gennes

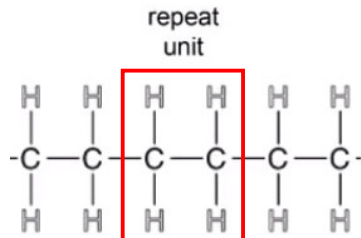
Nobel prize in Physics (1991)
For complex forms of matter,
liquid crystals and polymers



- P. Flory, *Principles of Polymer Chemistry*. Cornell University Press, 1954
- P. G.de Gennes; L. Leger, *Dynamics of Entangled Polymer Chains*. *Annu. Rev. Phys. Chem.* **1982**, 33 (1), 49–61
- M. Doi, *Introduction to polymer physics*, Oxford university press, 1996
- L. R. Treloar, *The Physics of Rubber Elasticity*. Oxford Univ. Press, 2005
- Treloar, L. R. (1974). The mechanics of rubber elasticity. In *Journal of Polymer Science: Polymer Symposia* (Vol. 48, No. 1, pp. 107-123). New York: Wiley Subscription Services, Inc., A Wiley Company.

Polymer Many Units

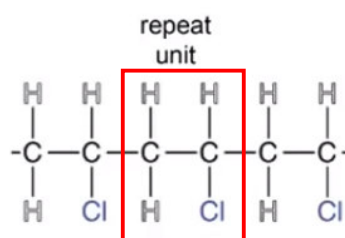
Definition:



Polyethylene(PE)



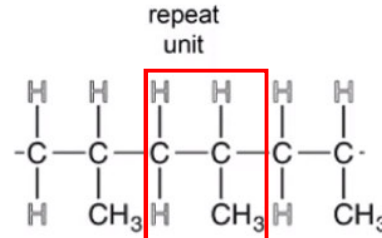
PE Milk Container



Poly(vinyl chloride) (PVC)



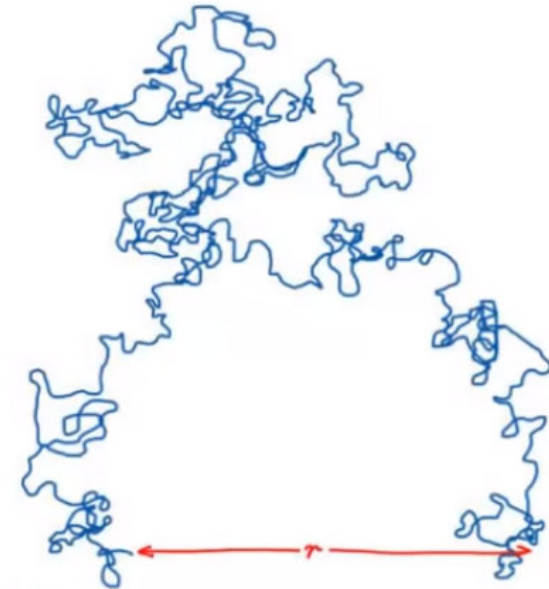
PVC Pipe



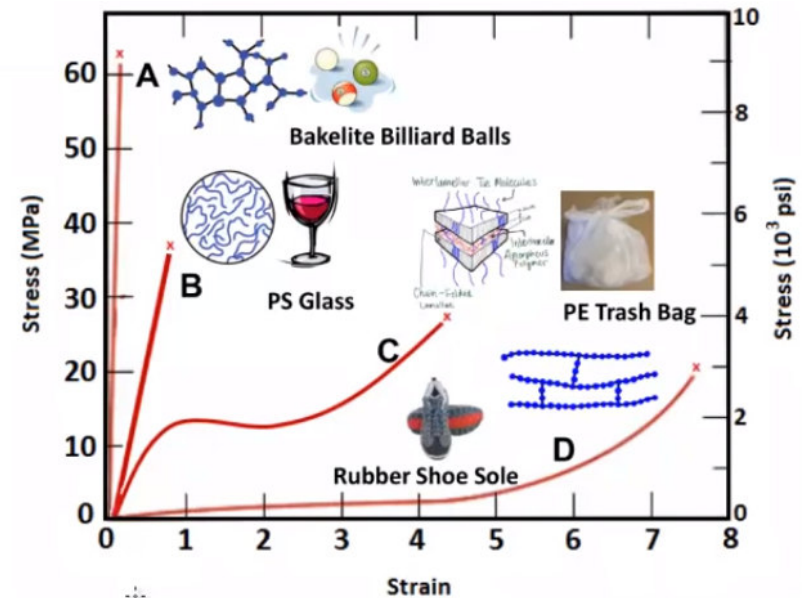
Polypropylene (PP)



PP Rope



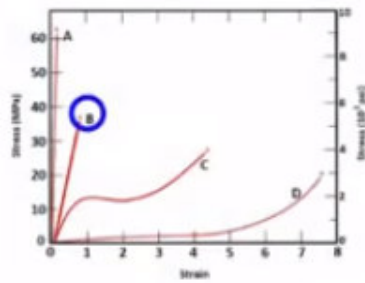
Polymer	E (MPa)	TS (MPa)	%EL
Natural Rubber	2.5	20	600-1000%
HDPE	830	28	300-600%
Polystyrene	3100	40	1.2-2.5%
Bakelite	6900	55	0.1%



Thermoplastic Polymers

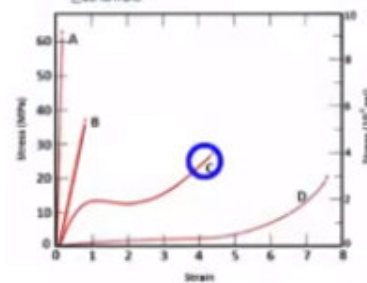
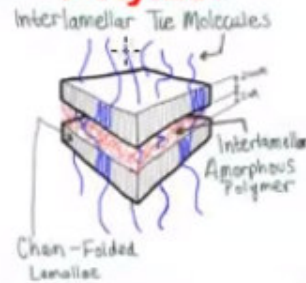
Heat Set Polymers

Amorphous Polymer



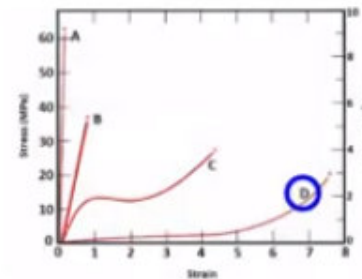
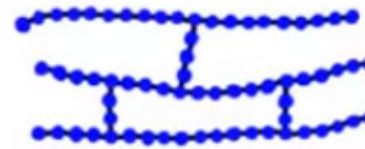
PS Glass
 $T_g = 100C$
 $T_{25} < T_g$
 Glassy

Semicrystalline Polymer



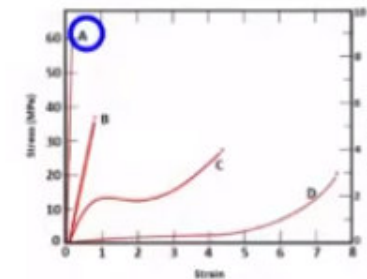
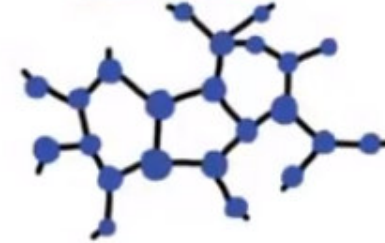
PE Trash Bag
 $T_g = -90C$; $T_m = 130C$
 $T_g < T_{25} < T_m$
 Flexible

Lightly X-linked Elastomer



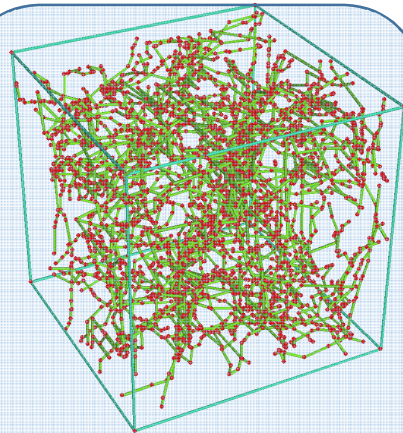
Rubber Shoe Sole
 $T_g = -103C$
 $T_g < T_{25}$
 Elastic

Heavily X-linked Thermoset

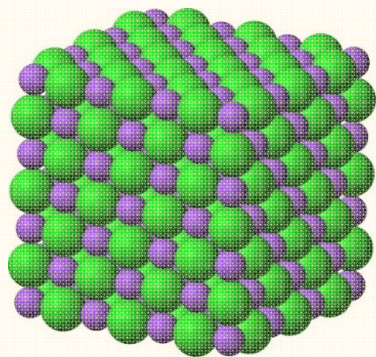


Bakelite Billiard Balls
 $T_g = 163C$
 $T_{25} < T_g$
 Rigid

T_g = Glass Transition Temperature, below which a polymer is rigid and brittle



*Polymer network:
entropic elasticity*



*Crystalline material:
enthalpic elasticity*

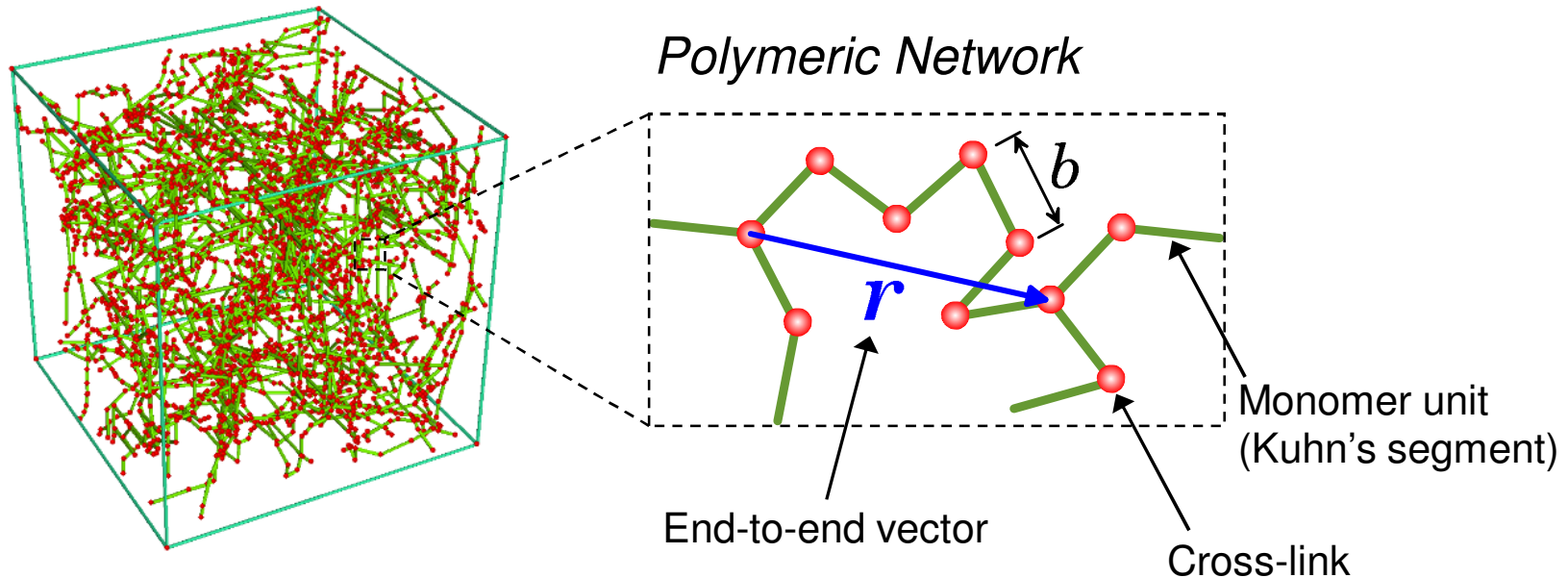
Polymer's structure:

Network of long molecular chains linked together.

The mechanical behaviour is ruled by **entropic energy ΔS**

Polymers have a **high fracture toughness** due to the **high deformability** because of the **alignment capability** of chains in the tensile direction.

Mechanical response greatly depends on the **strain rate** because of the time-dependent phenomena occurring in the polymeric network.



$$\begin{cases} \mathbf{r} & = \text{current end-to-end vector} \\ \mathbf{r}_0 & = \text{reference end-to-end vector} \\ \mathbf{F} & = \text{deformation gradient tensor} \end{cases}$$

$$\Psi(\mathbf{F}) = -T \cdot \Delta S = \frac{1}{2} k_B T [\text{tr}(\mathbf{F}^T \mathbf{F}) - 3]$$

Fundamental relations in the mechanics of polymers:

$$G = n K_B T \quad \text{Shear modulus}$$

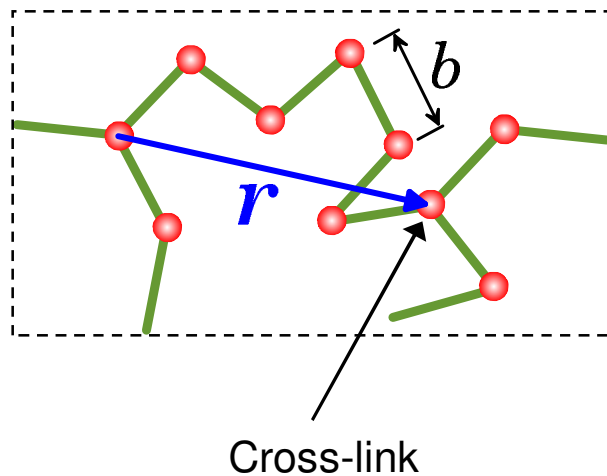
$$\nu = \sim 0.5 \quad \text{Polymers are nearly incompressible}$$

n = No. of chains per unit volume

K_B = Boltzmann constant

T = absolute temperature

Types of chain bonds:



- **Weak** physical bond
(ionic, dipole bonds, H bonds, van der Waals)

→ thermoplastic
polymers

- **Strong** (covalent bond)

→ thermoset elastomer (coarse bonds)

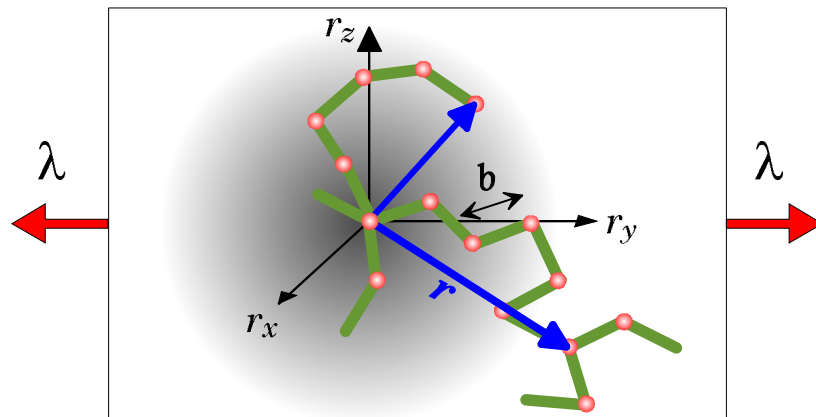
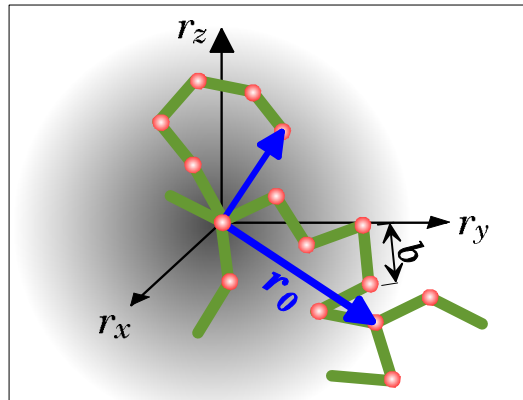
→ thermoset polymers (dense bonds)

- Bergstrom, J.S. (2015). *Mechanics of solid polymers: theory and computational modeling*. William Andrew.
- Fakirov, S. (2017). *Fundamentals of polymer science for engineers*. Wiley-VCH.
- Rajagopal, K.R., Wineman, A. S. (2001). *Mechanical Response of Polymers: an introduction*. Cambridge University Press

Chain and polymer deformation

Mean end-to-end distance

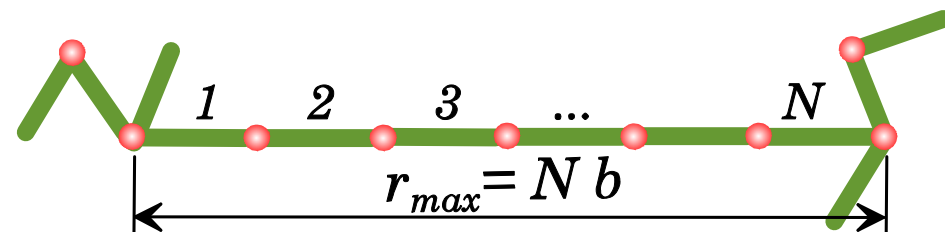
$$r_0 = b\sqrt{N}$$

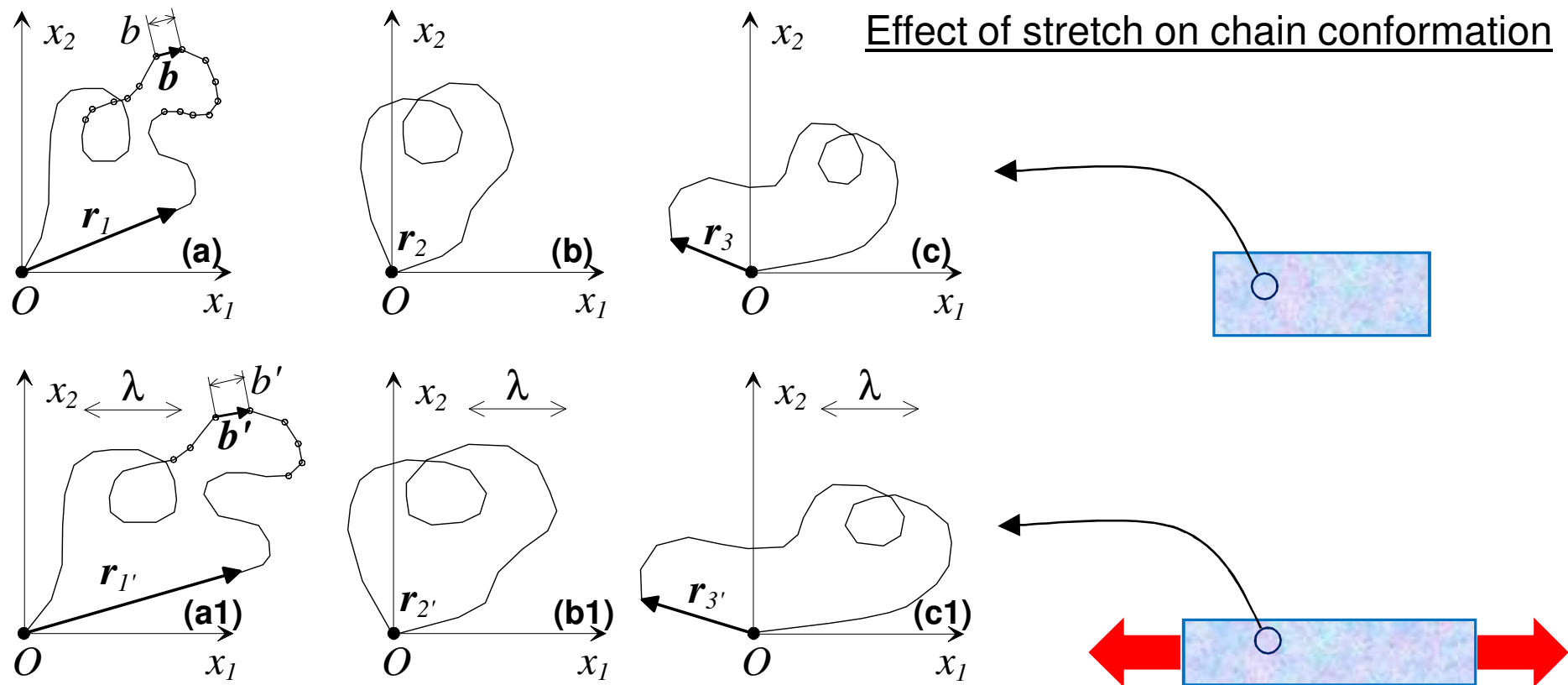


$$r = \lambda r_0 = \lambda b\sqrt{N}$$

$$r_{max} = bN \rightarrow \lambda_{max} = \sqrt{N}$$

Max allowable stretch



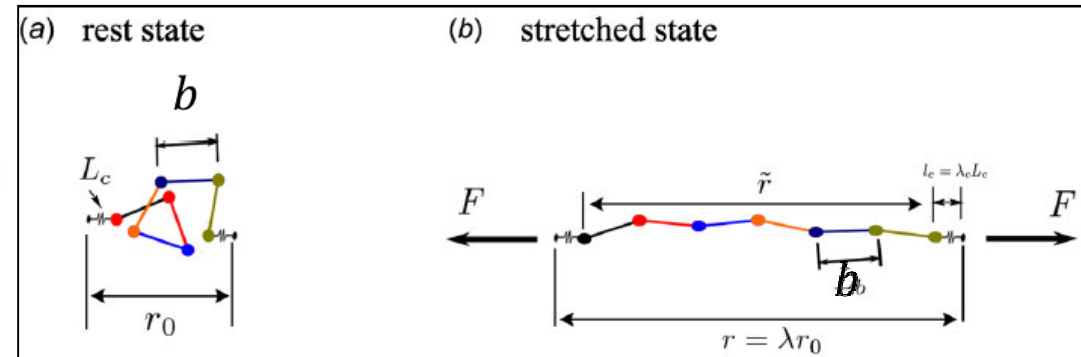
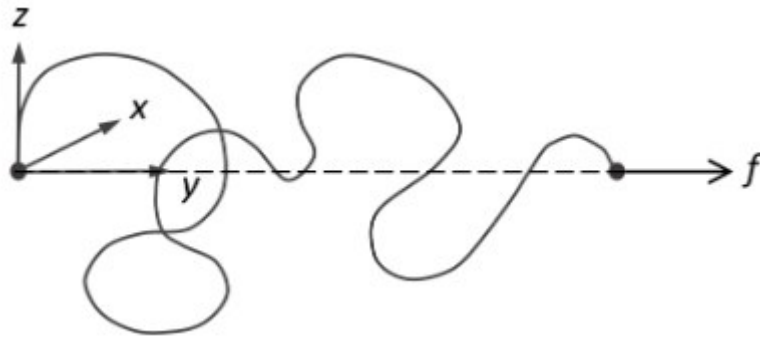


Work variation, related to the entropy of a single stretched chain:

$$\Psi(\mathbf{F}) = \frac{1}{2} k_B T \left[\text{tr}(\mathbf{F}^T \mathbf{F}) - 3 \right]$$

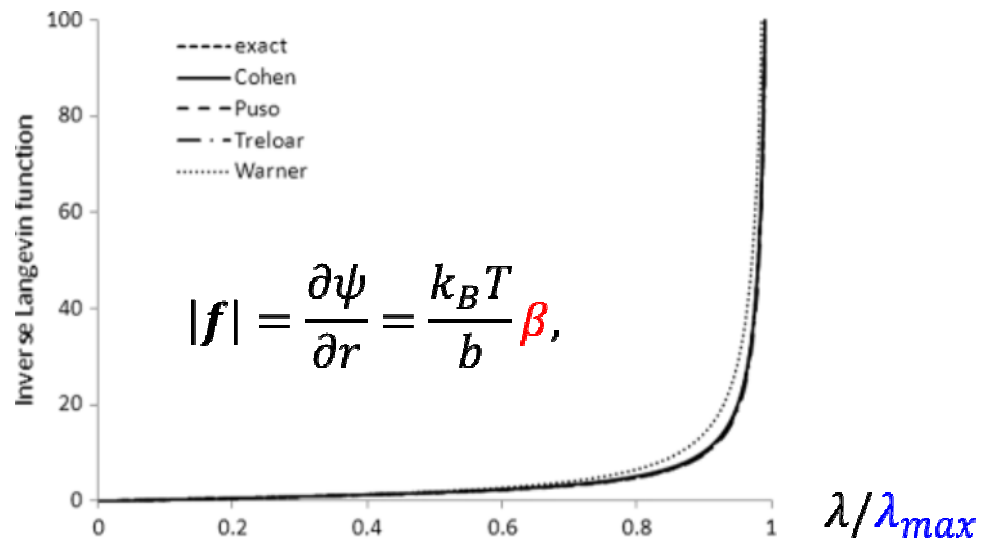
$$\Psi(\lambda) = -T \cdot \Delta S = \frac{1}{2} k_B T \left[\lambda_x^2 + \lambda_y^2 + \lambda_z^2 - 3 \right]$$

Chain stretch



$$\lambda_{max} = \sqrt{N}$$

Langevin statistics:



$$\psi(r) = k_B T \left(\frac{r}{bN} \beta + \ln \frac{\beta}{\sinh \beta} \right),$$

$$\beta = \mathcal{L}^{-1} \left(\frac{r}{bN} \right)$$

$$\mathcal{L}(x) = \coth(x) - x^{-1}$$

3. AM TECHNOLOGIES FOR POLYMERS

AM processes for polymers

CATEGORY	1	2	3	4	5	6
	powder bed fusion	vat photopolymerisation	material extrusion	material jetting	binder jetting	sheet lamination
DENOMINATIONS	SLS - Selective Laser Sintering SHS - Selective Heat Sintering MJF - Multi Jet Fusion	SLA - Stereolithography DLP - Digital Light Processing DLS - Digital Light Synthesis	FDM - Fused Deposition Modeling FFF - Fused Filament Fabrication	PolyJet MJM - Multi-Jet Modeling	3DP - 3D Printing	LOM - Laminated Object Manufacture SFP - Solid Foil Polymerisation
STATE	powder	liquid	solid	liquid	powder	solid
MATERIALS	polyamides, PEEK, polypropylene (PP), polycarbonate (PC), polystyrene (PS), thermoplastic elastomers (TPE)	epoxy or acrylic photopolymers	polylactic acid (PLA), ABS, PC	acrylic photopolymers	polycaprolactone (PCL), polyvinyl alcohol (PVA), PLA	PMMA, PVC
PRINCIPLE	selective fusion and solidification	light reactive photopolymer curing	extrusion of melted material and solidification	jetting of melted material and solidification / UV curing	consolidation through binder	lamination through thermal or chemical bonding




For a comprehensive review on AM of polymers see:

J Mater Sci (2021) 56:961–998

Review



Laser-based additively manufactured polymers: a review on processes and mechanical models

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²Department of Mechanics and Strength of Materials, University Politehnica Timisoara, Blvd. M. Viteazu, Nr. 1, 300222 Timisoara, Romania

182 up-to-date references!

Classification of AM categories according to ASTM

- 1) **powder bed fusion**
AM process in which thermal energy selectively fuses regions of a powder bed
- 2) **vat photopolymerization**
AM process in which liquid photopolymer in a vat is selectively cured by light-activated polymerization
- 3) **material extrusion**
AM process in which material is selectively dispensed through a nozzle or orific
- 4) **material jetting**
AM process in which droplets of build material are selectively deposited and solidified
- 5) **binder jetting**
AM process in which a liquid bonding agent is selectively deposited to join powder materials
- 6) **sheet lamination**
AM process in which sheets of material are bonded and selectively cutted to form a part

ISO/ASTM 52900:2015(E)



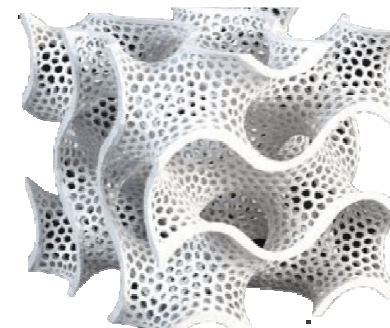
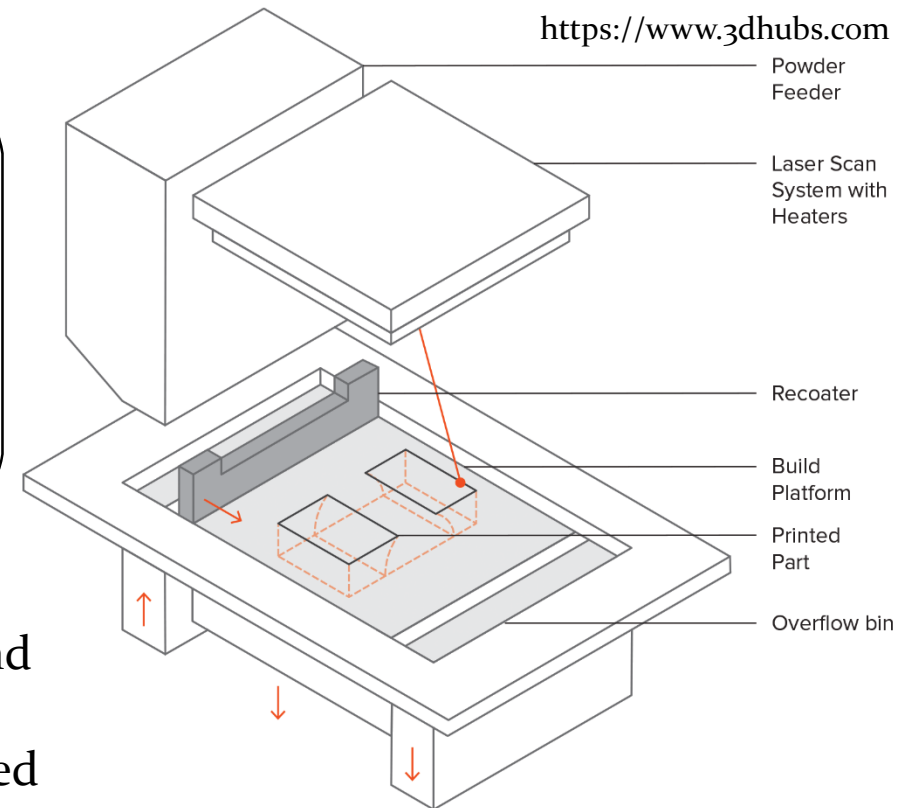
Standard Terminology for
Additive Manufacturing – General Principles –
Terminology^{1,2}

**6 categories for polymeric
materials**

1) Powder bed fusion

raw material form: **powder**
 principle: **selective fusion and solidification**
 resolution: **50-100 μm**
 printed materials:
polyamides, polystyrene, PEEK, thermoplastic elastomers

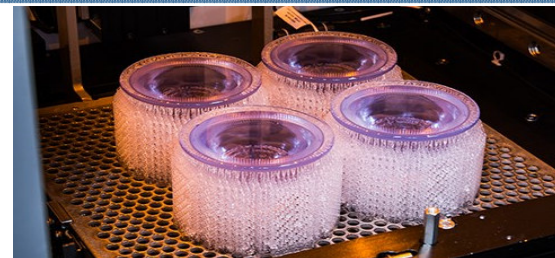
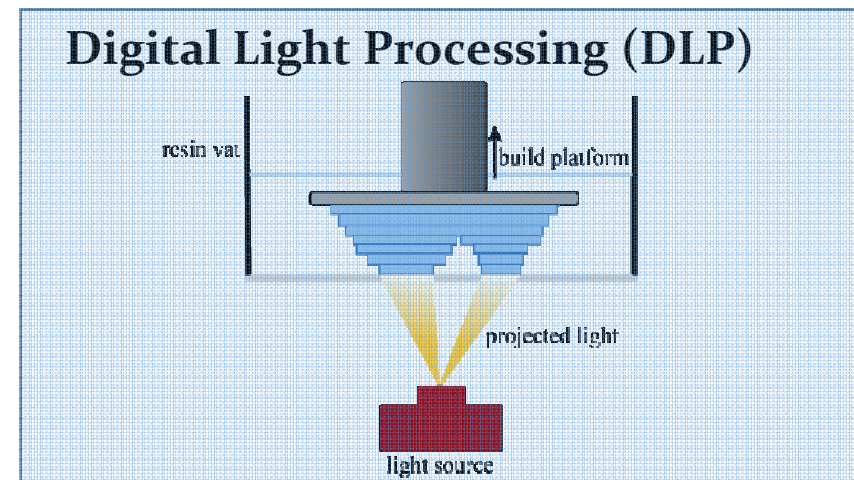
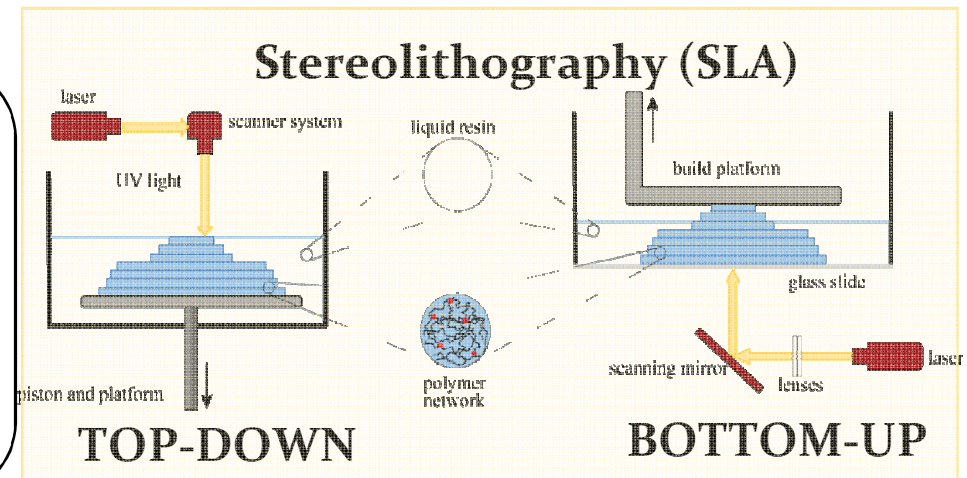
- comprises Selective Laser Sintering (SLS) and MultiJet Fusion
- particles are heated and then selectively fused by a moving laser scan
- **advantages:** good mechanical properties, low anisotropy
- **disadvantages:** porosity, rough surfaces



2) Vat photopolymerization

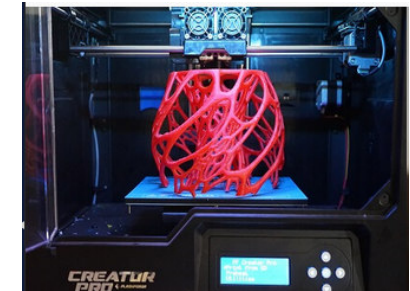
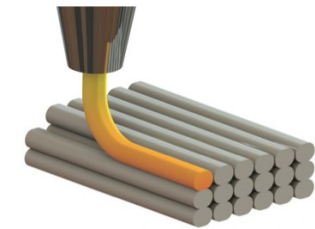
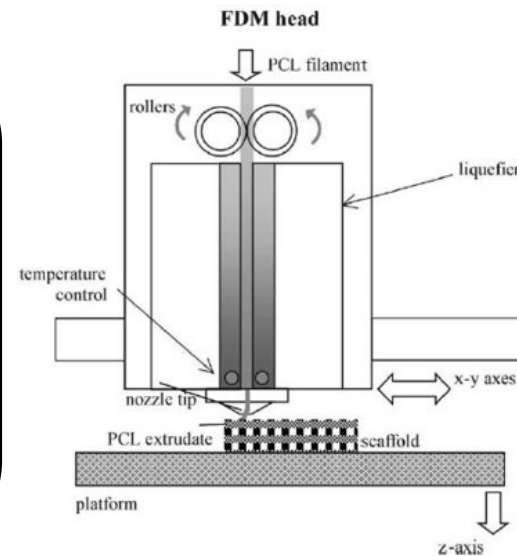
raw material form: **liquid**
 principle: **polymerization through light**
 resolution: **10-30 μm**
 printed materials:
 Epoxy or acrylic
 photopolymers

- a liquid resin initially placed in a vat is selectively irradiated and solidified (cured)
- based on two main technologies:
Stereolithography (a moving laser solidifies a layer) and Digital Light Processing (a layer is solidified in one shot)
- **advantages**: fast, extremely high resolution, allows multi-material AM
- **disadvantages**: post-treatment needed, expert users requiring, liquid resin manipulation



3) Material extrusion

raw material form: **solid**
principle: **extrusion of melted material and solidification**
resolution: **200 μm**
printed materials: **polylactic acid (PLA), acrylonitrile butadiene styrene (ABS), polycarbonate (PC)**



- the 3DP machine contains a plastic wire spool feeding a print head (nozzle) which extrudes thin filament of melted plastic
- mainly known with two acronyms (equivalent techniques): Fusion Deposition Modelling (FDM) and Fused Filament Fabrication (FFF)
- **advantages**: lower cost, accessibility (even house-machine)
- **disadvantages**: slow, poor resolution, high anisotropic components



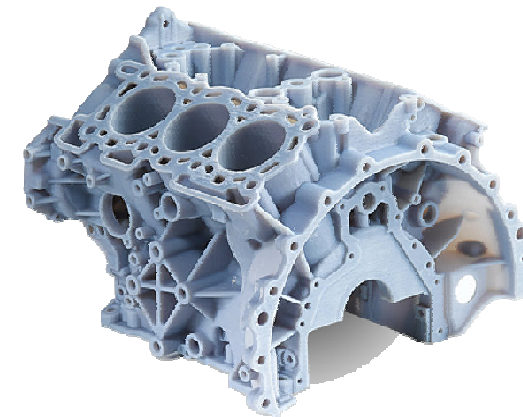
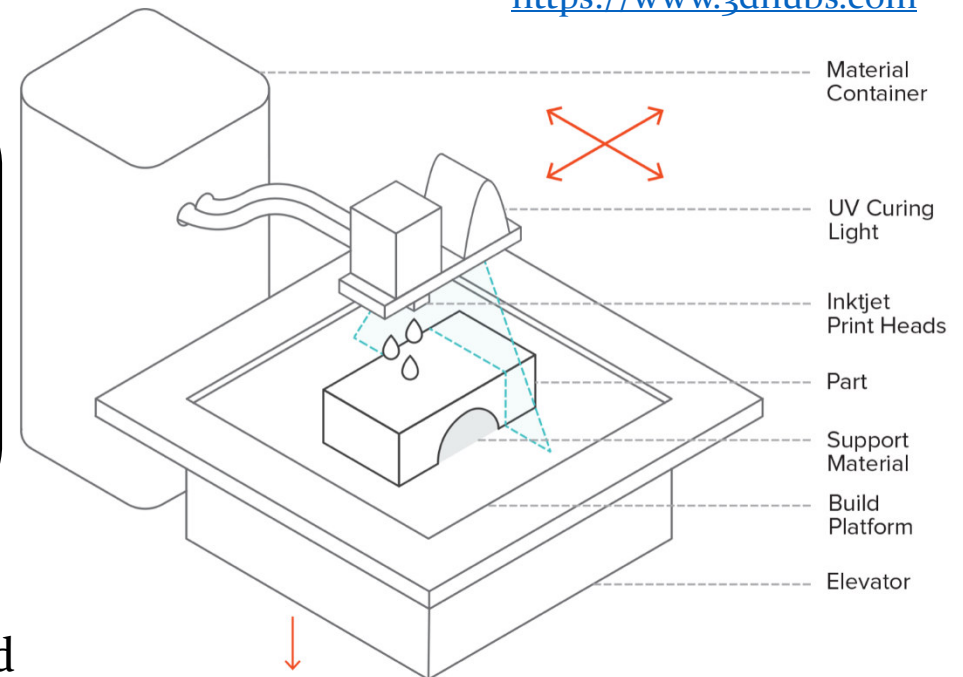
https://www.youtube.com/watch?v=ik39_sv-wgQ

4) Material jetting

raw material form: **liquid**
 principle: **thermal solidification or UV curing**
 resolution: **25 μm**
 printed materials:
acrylic photopolymers

- comprises thermal inkjet and inkjet-based lithography (PolyJet and MultiJet Modelling)
- heated droplets of photopolymer are deposited and cured by a UV light source
- advantages:** fast, allows multi-material AM, high accuracy, homogeneous materials
- disadvantages:** limited strength, photosensitive parts, high cost

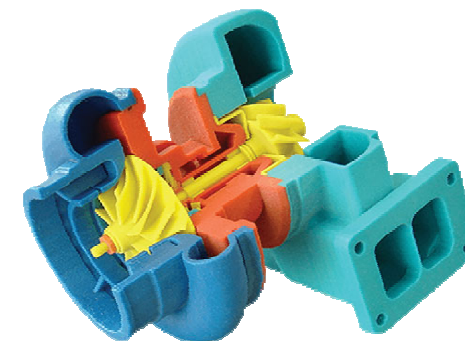
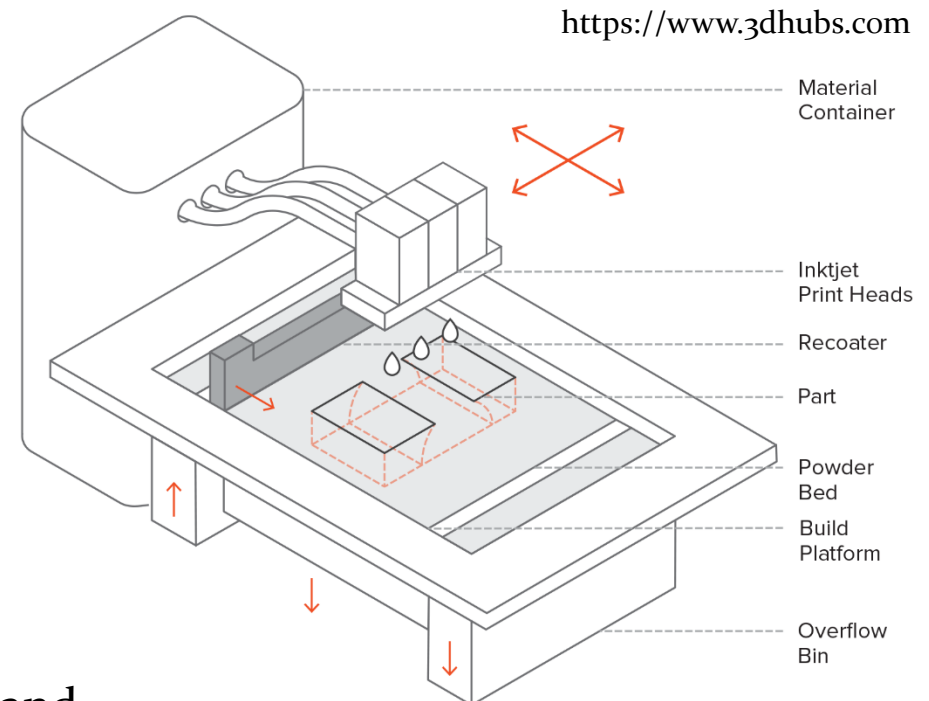
<https://www.3dhubs.com>



5) Binder jetting

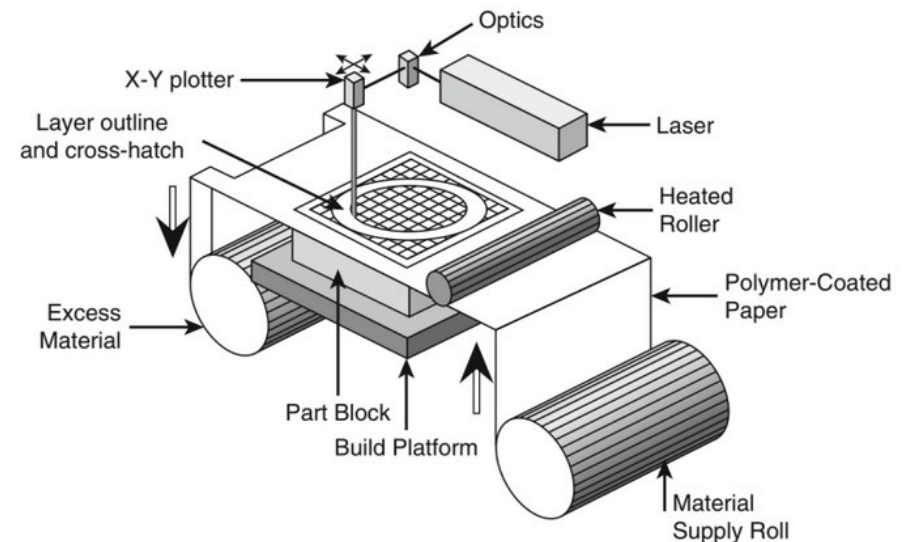
raw material form: **powder**
 principle: **consolidation through binder**
 resolution: **100 μm**
 printed materials:
**polylactic acid, polycaprolactone,
 polyvinyl alcohol, PMMA**

- also known as 3D printing
- powder layers are spread over the platform and a liquid ink then bonds or fuses the particles
- **advantages:** fast, allows multi-material AM, room temperature binding, no support
- **disadvantages:** limited strength, rough surfaces, limited material selection



6) Sheet lamination

raw material form: **solid**
principle: **deposition of sheet material with adhesives and successive laser cutting**
resolution: **200-300 μm**
printed materials: **actually not enough diffused in polymers (more metals)**



- material bonded with adhesive layer-by-layers and cutted with the required shape
- two basic technologies: laminated object manufacturing (LOM) and ultrasonic additive manufacturing (UAM)
- **advantages**: low temperature, no phase change, no support needed
- **disadvantages**: not so widespread, limited materials, poor resolution, high anisotropic components, waste of material



4. AM PROCESS PARAMETERS IN POLYMERS

To print a certain component with one of the previously shown AM techniques, we have to correctly set the process parameters.

Aim of design in AM:

1. **Choice of the AM process to print the desired component;**
2. **Decision of the optimum combination of the process parameters (of those involved in the chosen AM process) which provide a component with the desired characteristics**

depending by the final application of the component itself

- Load bearing capacity (mechanical properties);
- Surface quality;
- Hardness and wear performance;
- Designed porosity (example in bio-printing, scaffolds etc.);

Here we focus on the mechanical properties

How mechanical properties of the final AM material are influenced by the raw material used and by the printing parameters?

Issue: how mechanical properties of the printed component are related to AM process parameters?

The majority of the existing studies

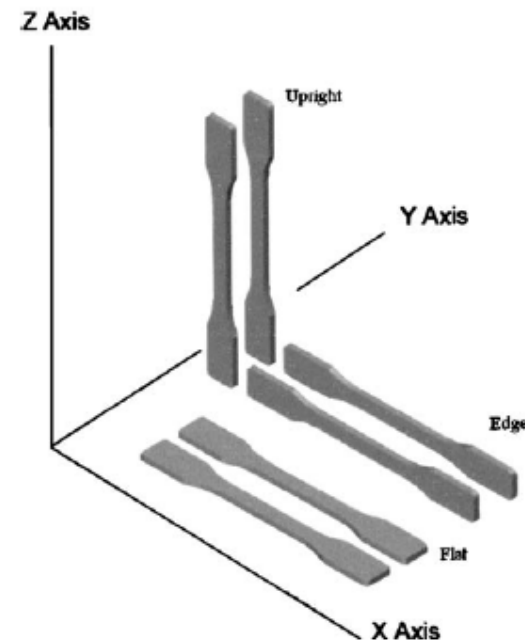
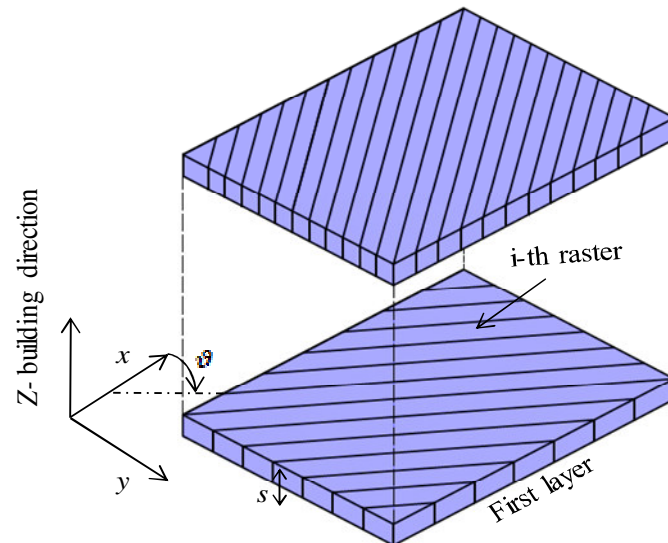
- Experimental approaches or statistical simulation-based
 - Help to visualize AM parameters – mechanical properties trend
 - Fail to accurately predict mechanical properties
 - Not able to explain the reason of such differences

Recently proposed studies

- Engineering approaches based on a mathematical formulation
 - Should provide quantitative prediction of the mechanics behaviour
 - Able to explain the understanding of a certain mechanical behavior in relation to a specific process parameter to improve performance of AM component in a more targeted way
 - “Tool” in the office for rapid prediction (experimental tests reduction)

AM process parameters

- Common to different AM processes
 - Layer thickness
 - Raster orientation
 - Building orientation
- Specific for an AM process, examples:
 - Nozzle temperature (FDM)
 - Laser energy irradiated (SLA and SLS)
 - Powder particle size (SLS)

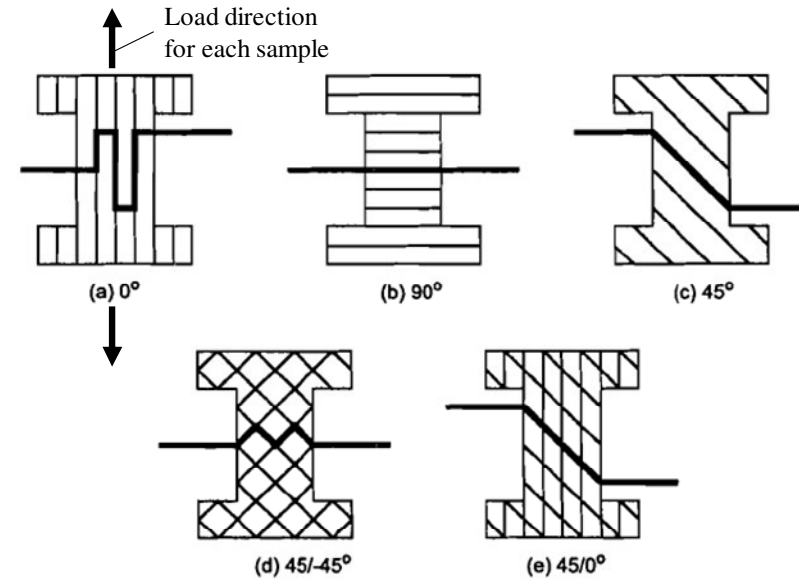


Fusion Deposition Modeling (FDM)

- Raster orientation

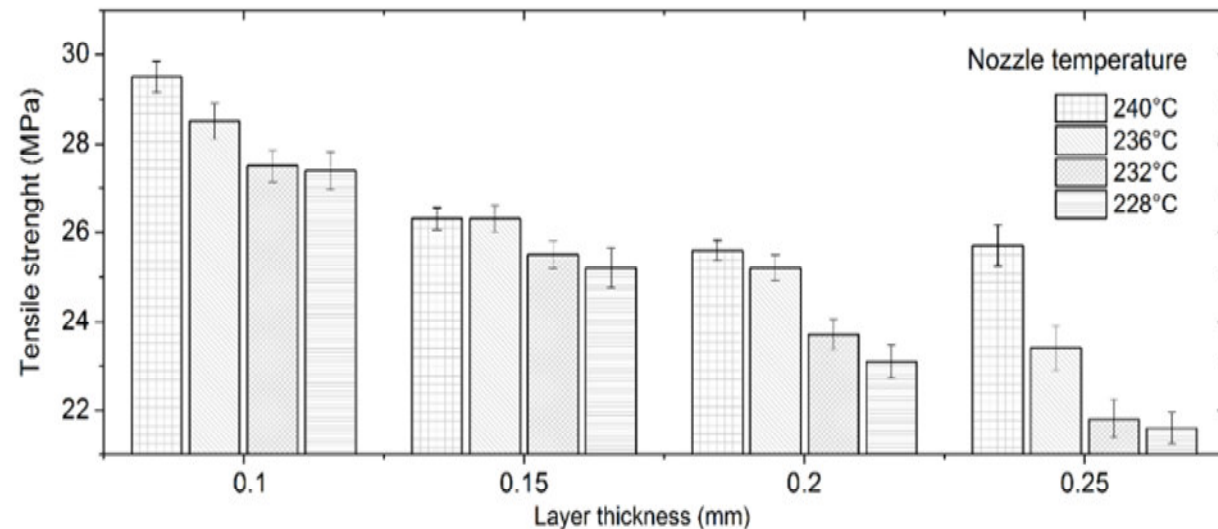
Schematic representation of rupture geometry on ABS samples printed with FDM process by varying only raster orientation

Es-Said, O. S. *Materials and Manufacturing Processes* 2000, 15(1), 107-122



- Layer thickness
- Nozzle temperature

Maloch, J. In: *Materials Science Forum* 2018 (Vol. 919, pp. 230-235).



Binder jetting (3D printing)

- powder properties
 - particle diameter ($d > 5 \mu\text{m}$)
 - particle shape (spherical is better)
- binder
 - type (aqueous vs non-aqueous ink)
 - saturation
 - properties (viscosity, surface tension)
- post-processing treatments
 - infiltrations with resins
 - sintering or cold pressing



large particles ensure easier spreading



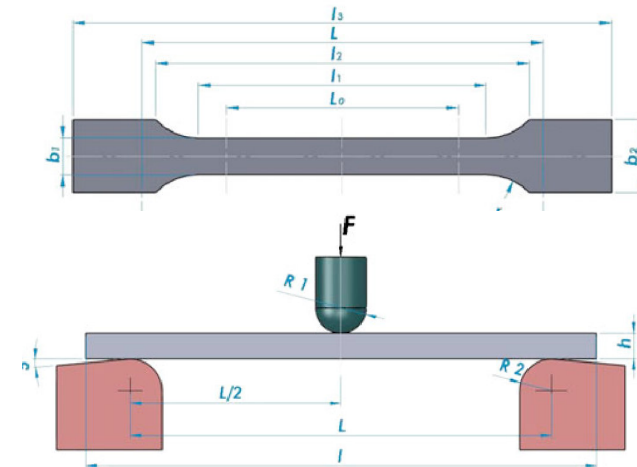
spreading, pore size,
surface roughness,
resolution



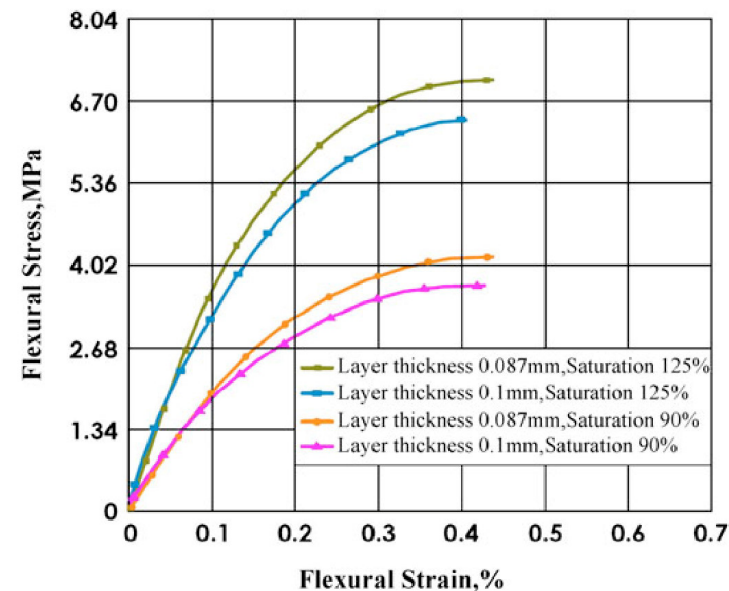
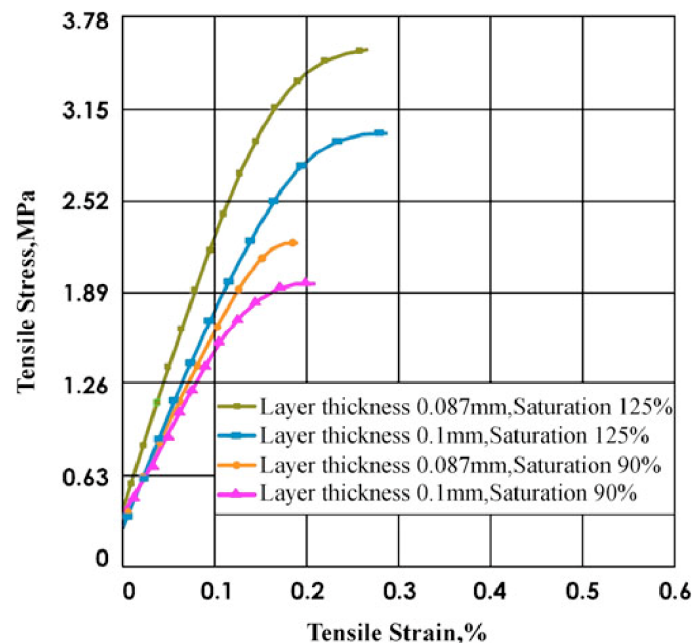
small particles ensure
increased resolution and
reduced roughness

Binder jetting (3D printing)

- effect of **increased** layer thickness:
 - better surface uniformity
 - reduced tensile strength
 - decreased flexural strength
- effect of **increased** binder saturation:
 - increased tensile and flexural strength

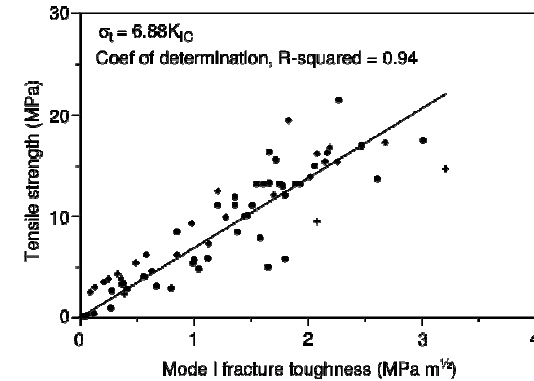
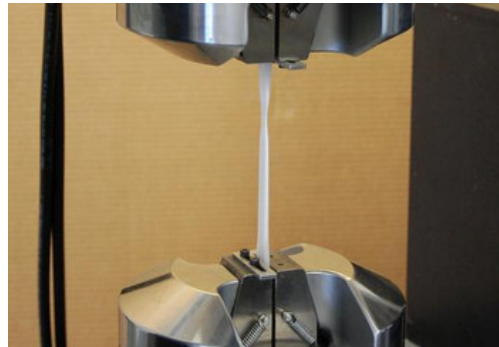
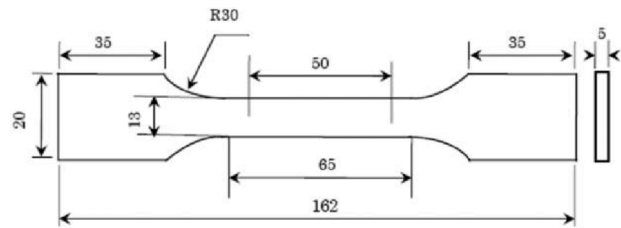


Vaezi, M. *Int. J. Adv. Manuf. Technol.* **2011**, 53, 275-284

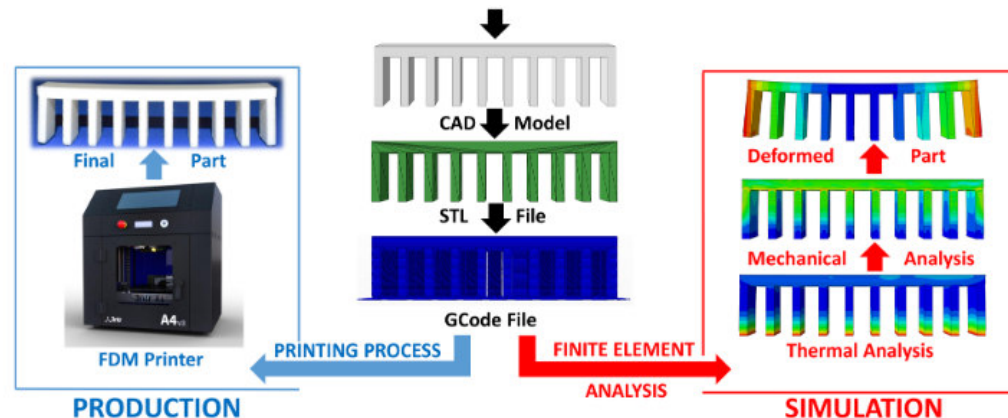
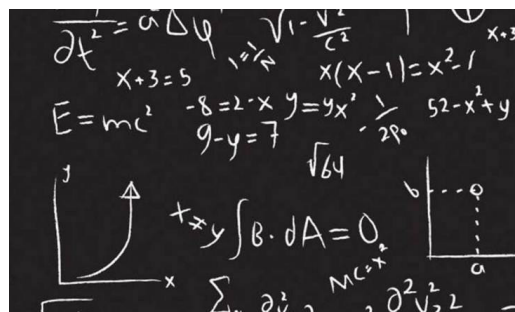
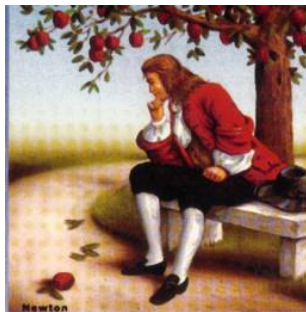


ENGINEERING APPROACHES TO THE DESIGN OF LASER-BASED AM POLYMERIC MATERIALS

1) Experiments

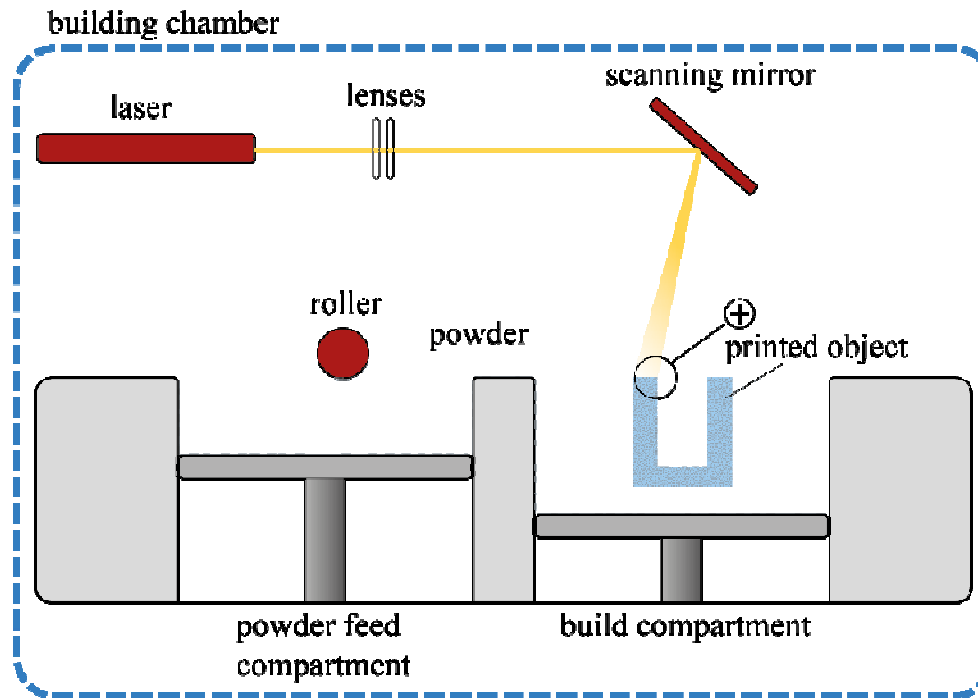
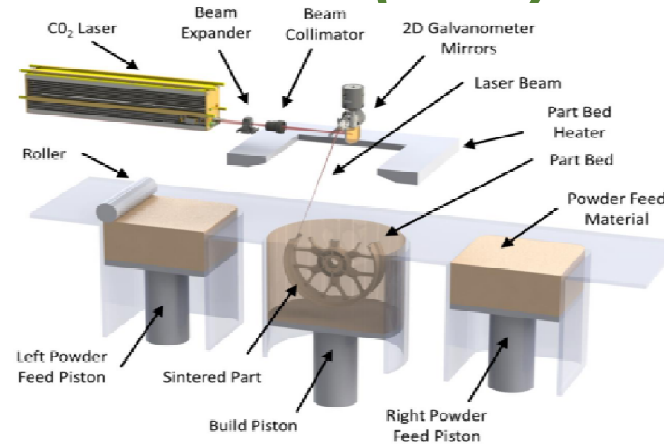


2) Simulations

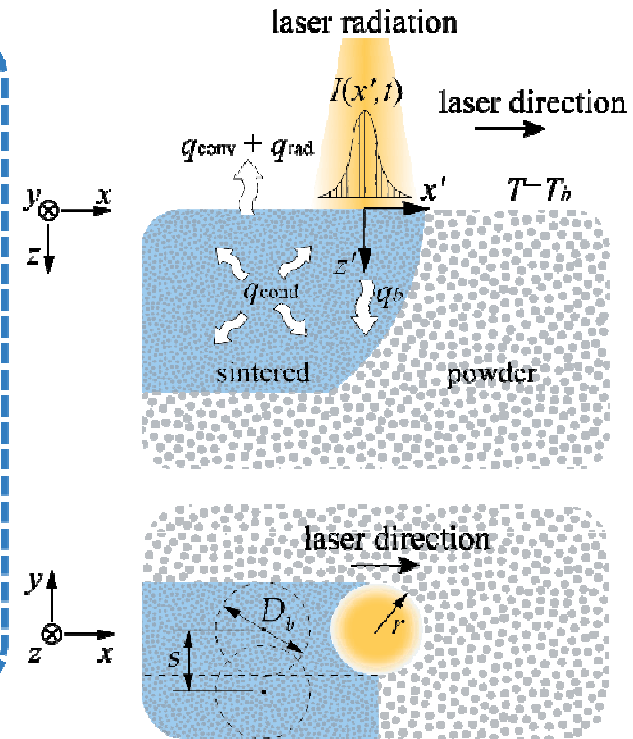


A brief illustration on these approaches is reported in the following slides for as concerning selective laser sintering (SLS) and photopolymerization (SLA) processes

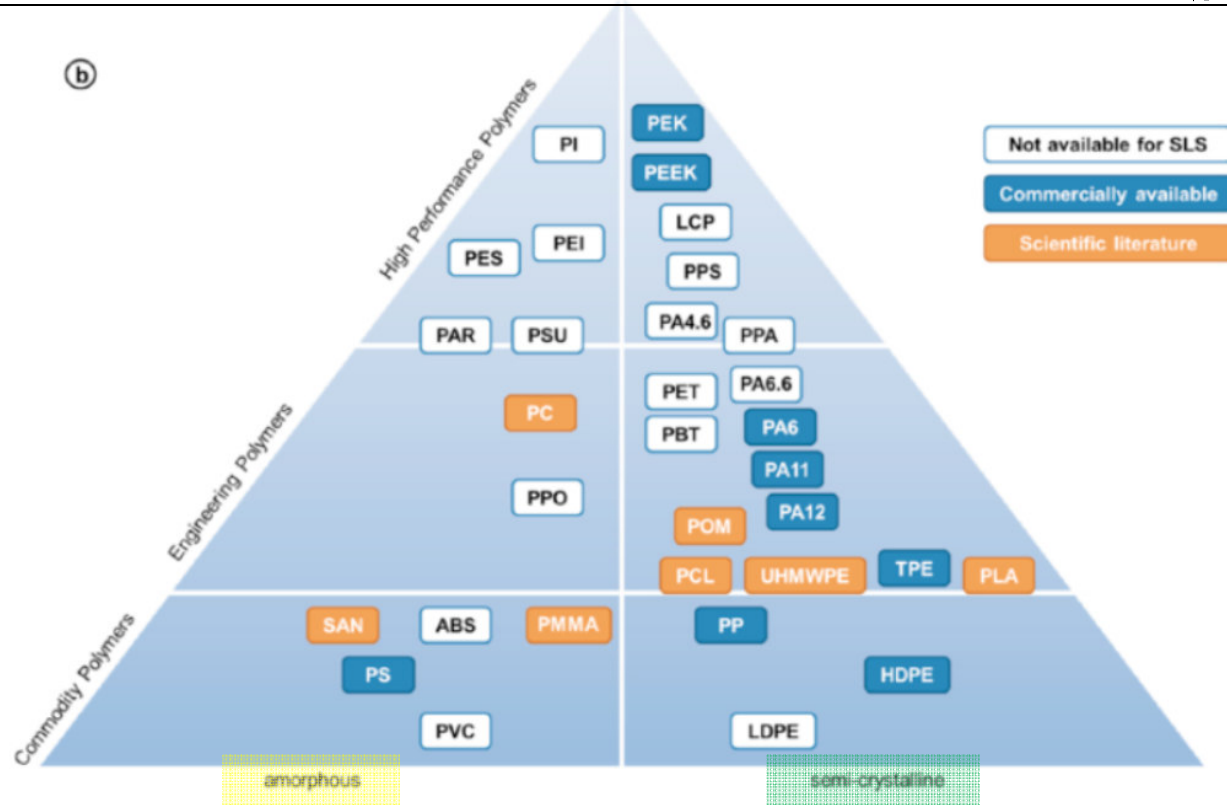
5. SELECTIVE LASER SINTERING (SLS): EXP/SIMULATIONS



(a)



(b)



Amorphous

PS: Polystyrene; SAN: Styrene acrylonitrile; PMMA: Poly-methyl methacrylate; PC: Polycarbonate.

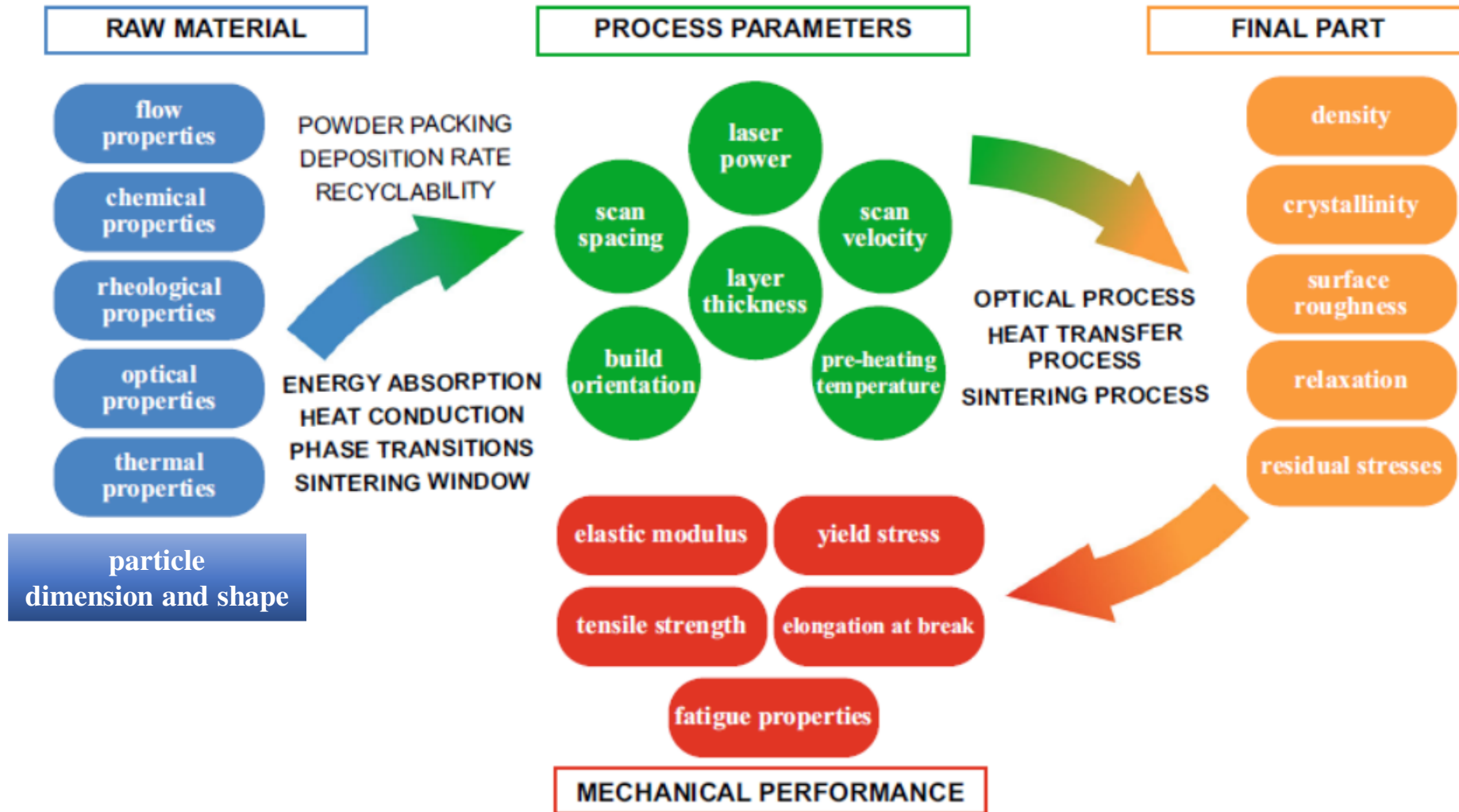
Not available for LS: PVC: Polyvinyl chloride; ABS: Acrylonitrile Butadiene Styrene; PPO: Polyphenylene oxide; PSU: Polysulfone; PES: Polyethersulfone; PEI: Polyetherimide; PI: Polyimide.

Semicrystalline

HDPE: High Density Polyethylene; PP: Polypropylene; PA_n : Polyamide- n ; PLA: Polylactic acid; UHMWPE: Ultra-high-molecular-weight polyethylene; PCL: Polycaprolactone; POM: Polyoxymethylene; PEEK: Polyether ether ketone; PEK: Polyetherketone.

Not available for LS: LDPE: Low Density Polyethylene; PBT: Polybutylene terephthalate; PET: Polyethylene terephthalate; PPA: Polyphthalamide; PPS: Polyphenylene sulfide; LCP: Liquid-crystal polymers.

1) Experiments



1) Experiments

Definition of energy density

$$E_s = \frac{P}{sv}$$
 surface energy density E_s = laser power P / (scan spacing s * scan velocity v)

or

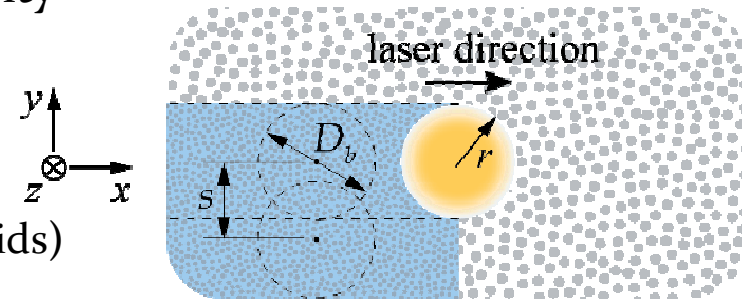
$$E_s = \frac{P}{sv} OL$$
 surface energy density E_s = (laser power P / (scan spacing s * scan velocity v)) * overlay $OL = D_b/s$

$$E_v = \frac{P}{svd}$$
 volumetric energy density E_v = laser power P / (scan spacing s * scan velocity v * layer thickness d)

Pilipović, A. *Adv. Mech. Eng.* 2014, 6(1), 648562

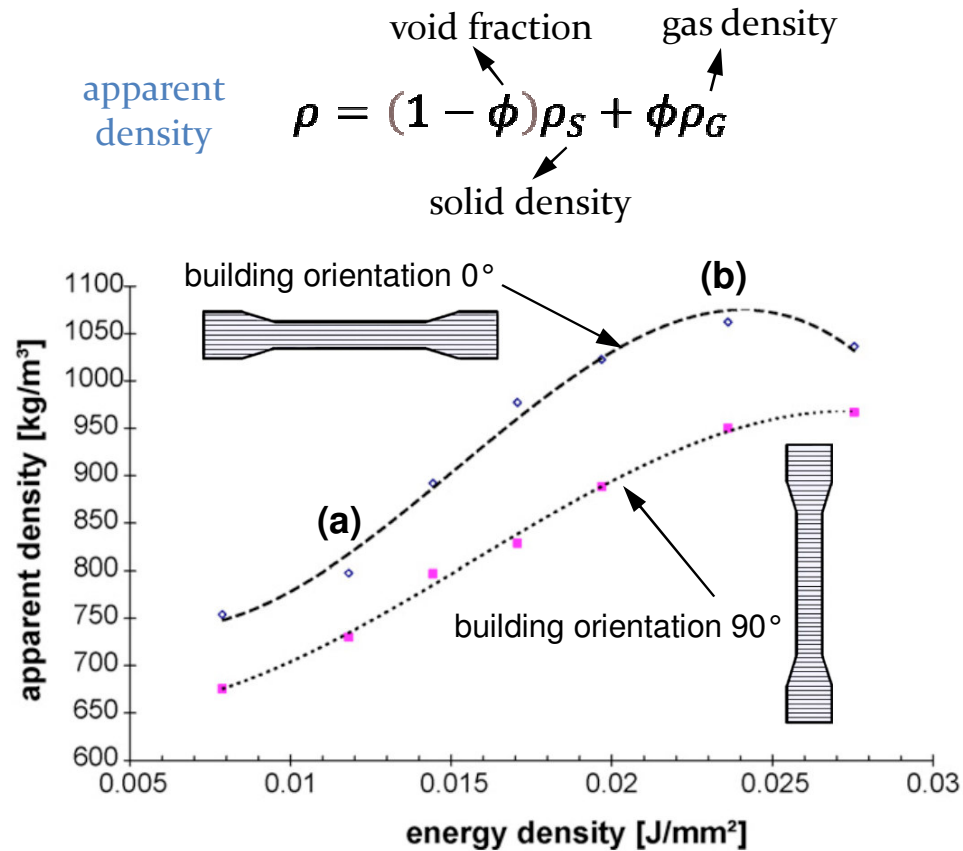
- positive correlation of energy density with density and mechanical properties

- optimal energy density \Rightarrow increased part density
- low energy density \Rightarrow incomplete sintering (voids)
- high energy density \Rightarrow thermal degradation



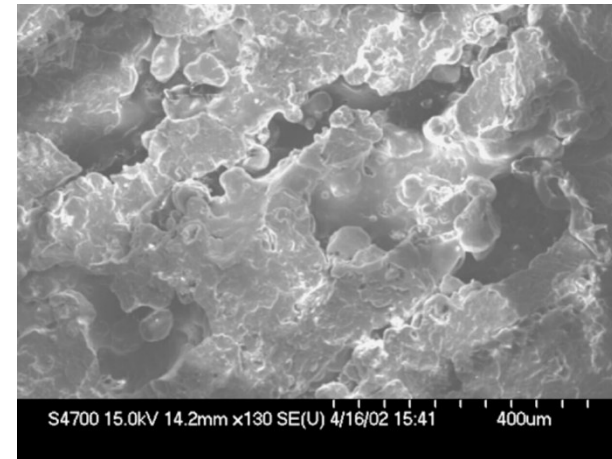
1) Experiments

Effect of energy density on **part density**

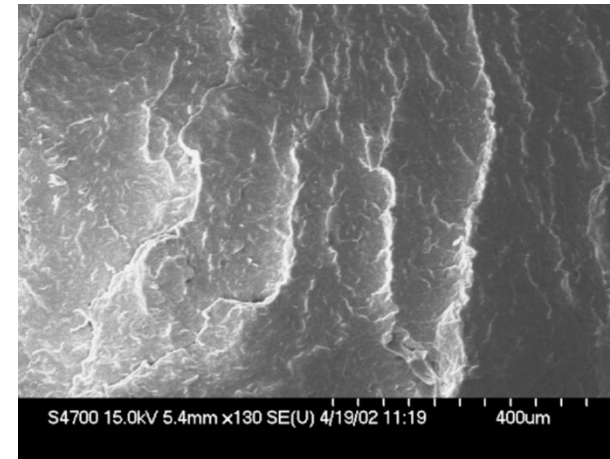


material: polyamide 12 (DuraForm PA)
 scan spacing $s = 0.15$ mm
 scan velocity $v = 5080$ mm/s

(a) $E_s = 0.012$ J mm⁻²



(b) $E_s = 0.024$ J mm⁻²



1) Experiments

Effect of energy density on **mechanical behaviour**

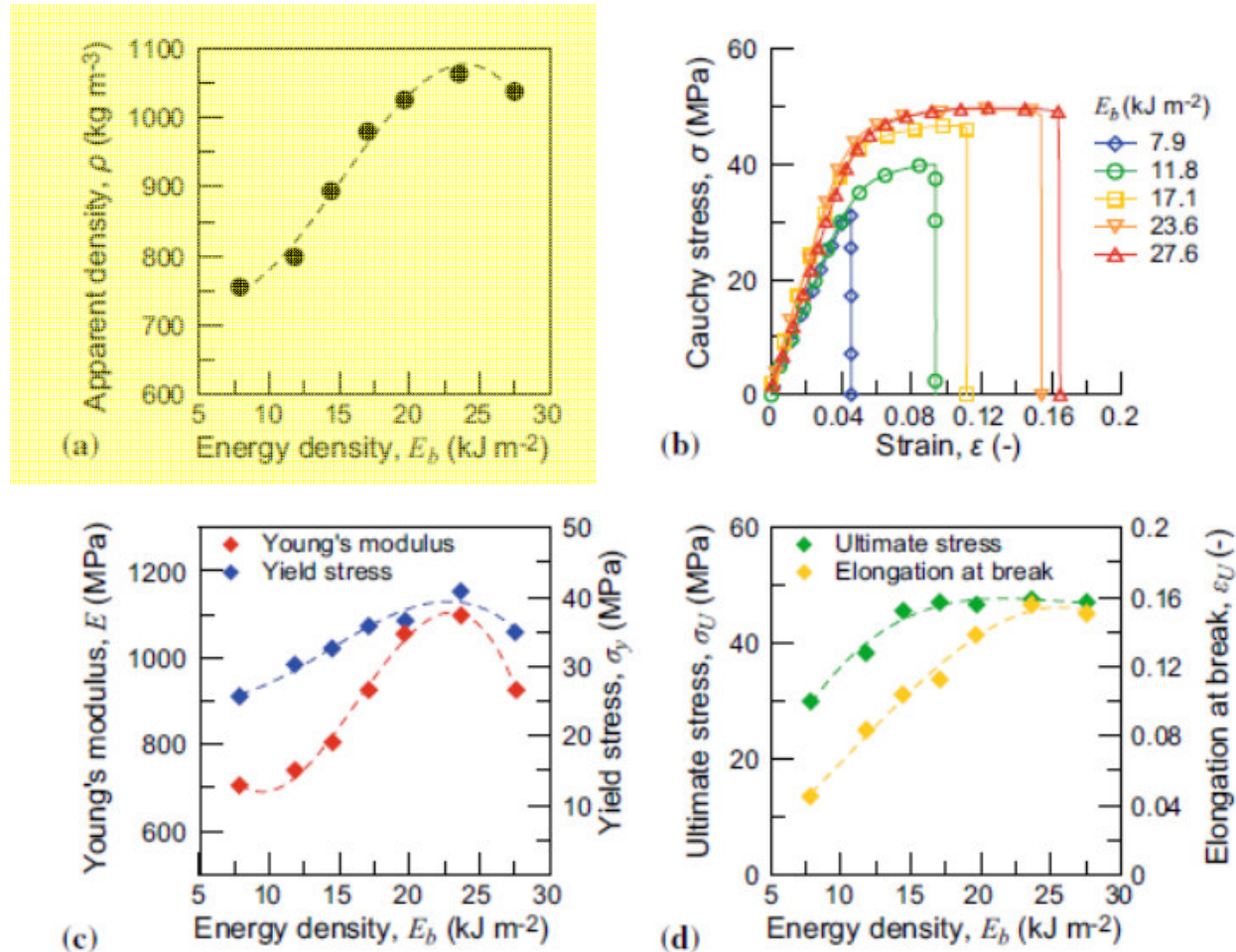
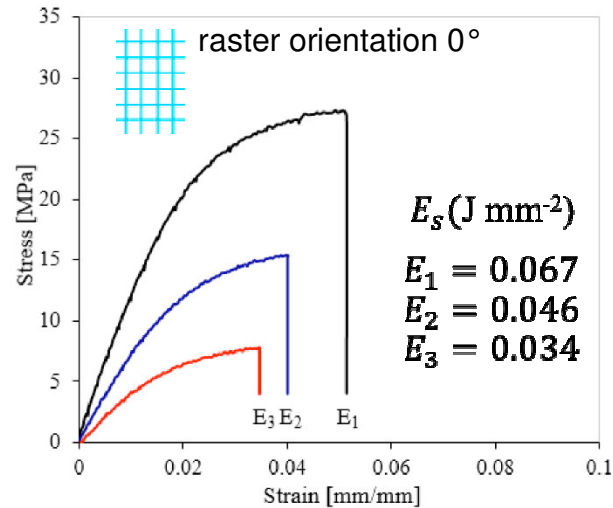


Figure 3 Dependence of the mechanical properties of printed samples on the surface energy density E_b . **a** Apparent part density. **b** Stress–strain curves. **c** Young’s modulus and yield stress. **d**

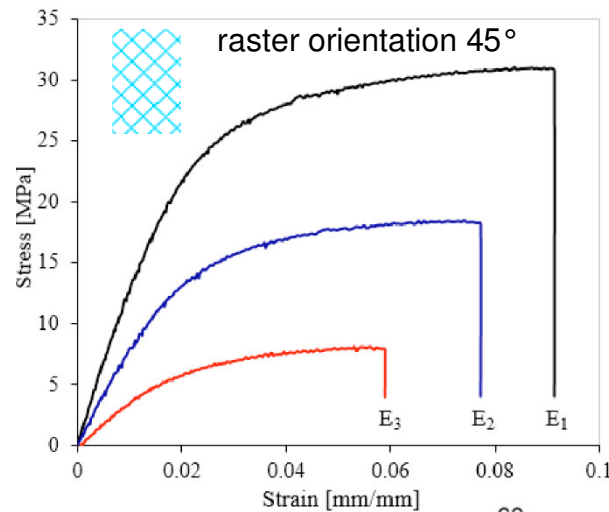
Ultimate stress and elongation at break (from uniaxial tension experiments, parallel to build orientation). Material is PA12 ($P = \text{var.}$, $v_b = 5.1 \text{ ms}^{-1}$, $s = 150 \mu\text{m}$). Adapted from [48].

1) Experiments

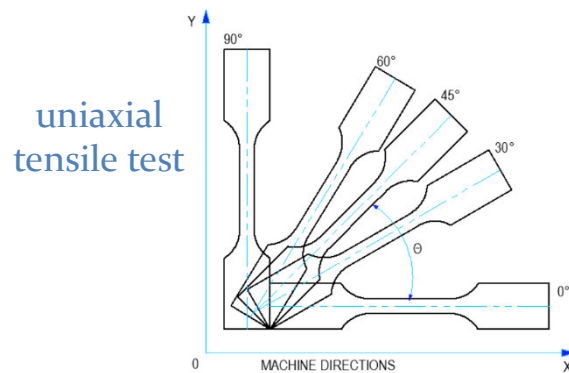
Effect of energy density on **mechanical behaviour**



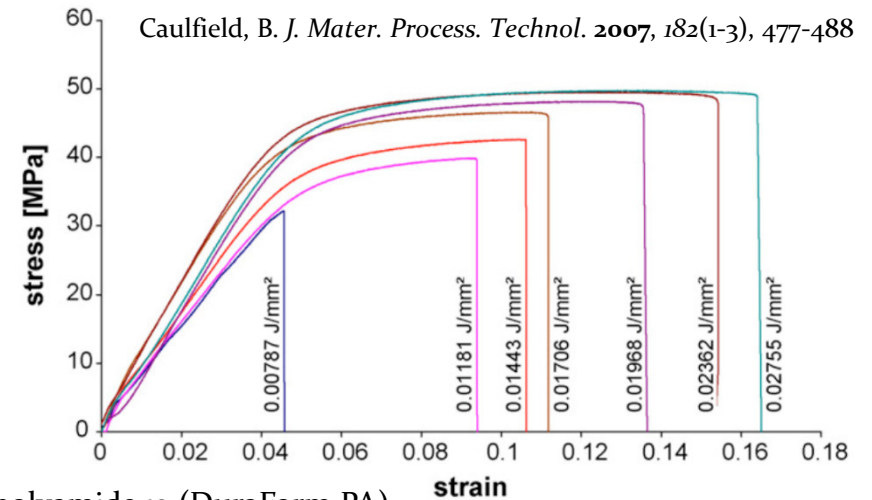
Stoia, D.I. *Polymers*. **2019**, 11(11), 1850



- in general, we observe a positive correlation between energy density and the mechanical behaviour, in terms of increased **stiffness** and **ductility**



material: polyamide 12 (PA2200)
scan spacing $s = 0.25$ mm
scan velocity $v = 1500$ -2500 mm/s



material: polyamide 12 (DuraForm PA)
scan spacing $s = 0.15$ mm
scan velocity $v = 5080$ mm/s

Caulfield, B. J. *Mater. Process. Technol.* **2007**, 182(1-3), 477-488

1) Experiments

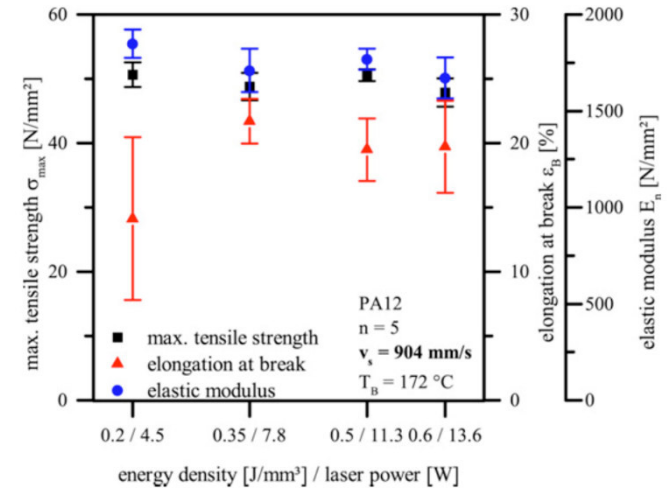
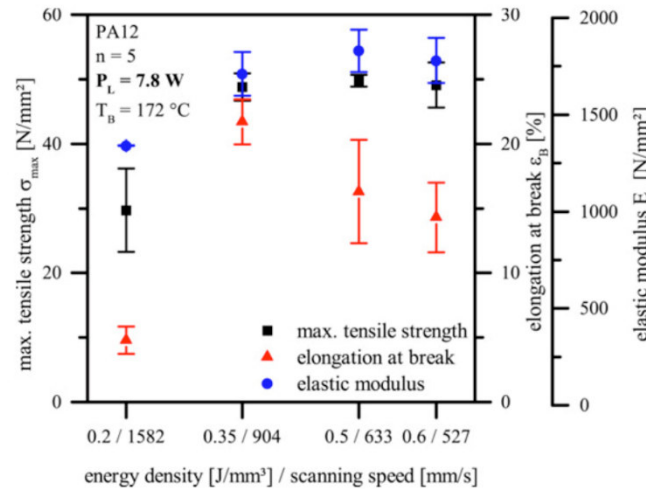
Effect of energy density on ultimate properties

Drummer, D. *Phys. Procedia*. 2014, 56, 176-183

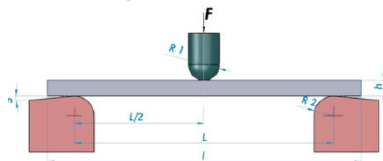
uniaxial tensile test



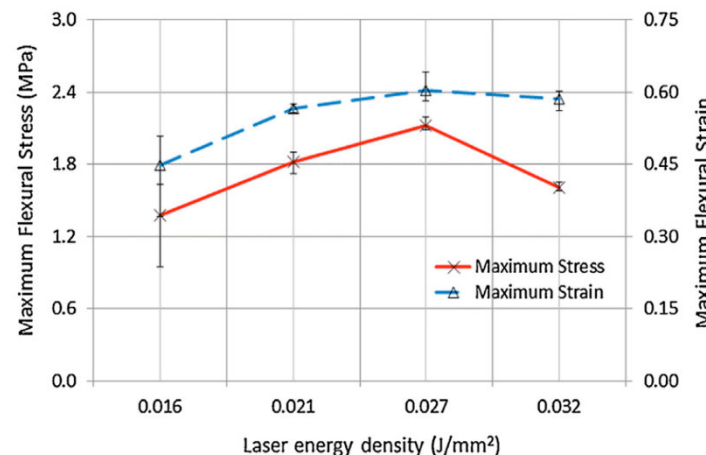
material: polyamide 12 (PA2200)
 scan spacing $s = 0.25$ mm
 scan velocity $v = 527-1582$ mm/s



three-point bending test



material: UHMWPE
 scan spacing $s = 0.15$ mm
 scan velocity $v = 2500$ mm/s



- ultimate properties, such as **maximum tensile strength**, **elongation at break** and **flexural strength** are deeply affected by energy density because of the different part density

Khalil, Y. *Addit. Manuf.* 2016, 10, 67-75

1) Experiments

Effect of raster orientation on **mechanical properties**

$$E_s (\text{J mm}^{-2})$$

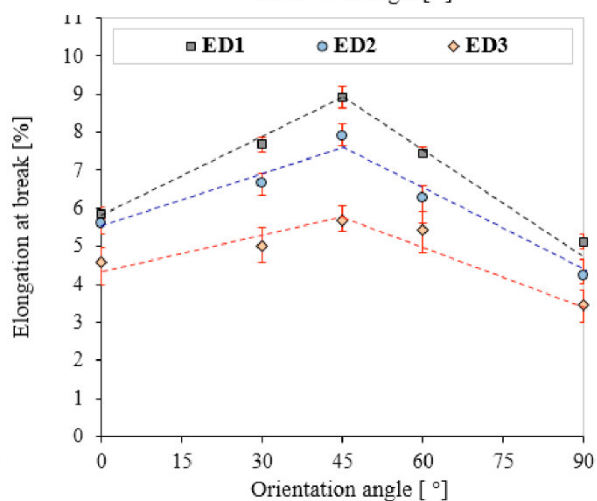
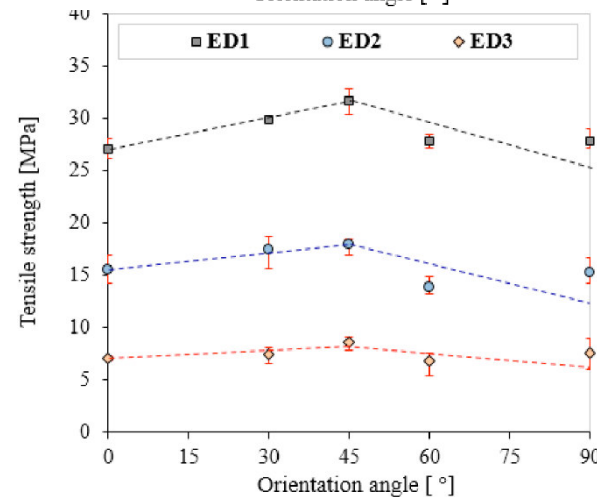
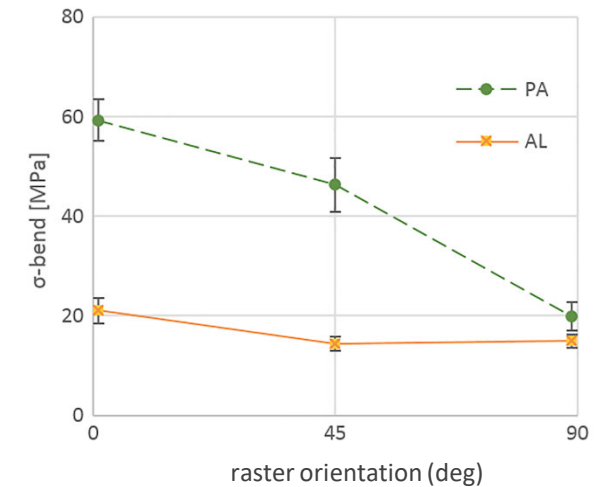
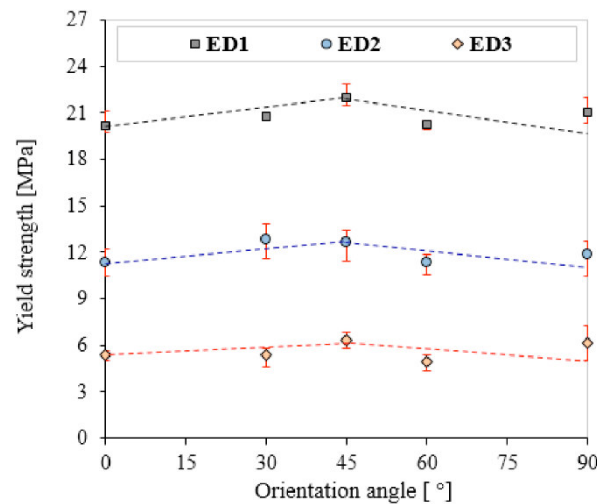
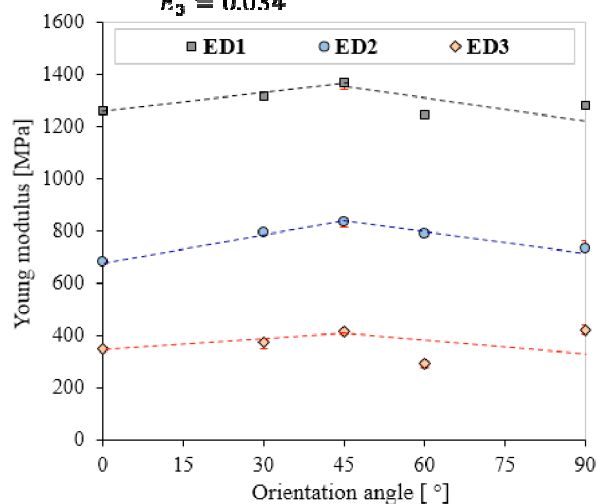
$$E_1 = 0.067$$

$$E_2 = 0.046$$

$$E_3 = 0.034$$

uniaxial
tensile test

three-point
bending test



Marsavina, L. *Mater. Des. Process. Commun.* **2020**, *2*(1), 1-5

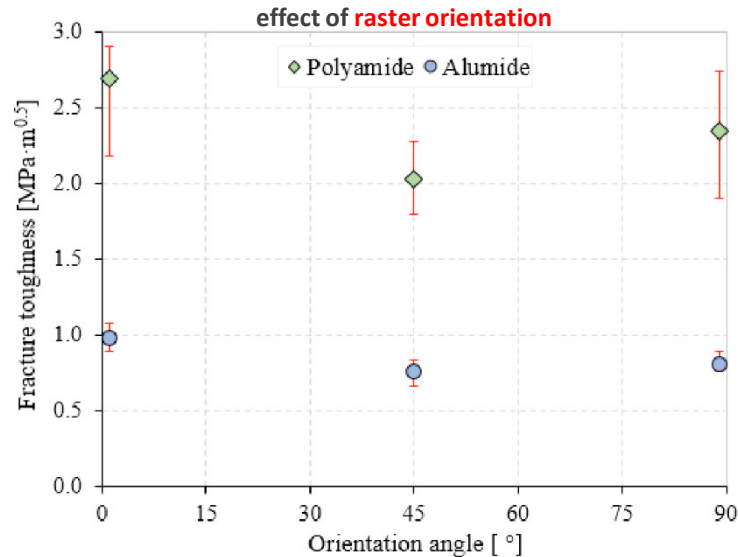
material: polyamide 12 (PA2200)
scan spacing $s = 0.25$ mm
scan velocity $v = 2000$ mm/s

Stoia, D.I. *Polymers.* **2019**, *11*(11), 1850

material: polyamide 12 (PA2200)
scan spacing $s = 0.25$ mm
scan velocity $v = 1500-2500$ mm/s

1) Experiments

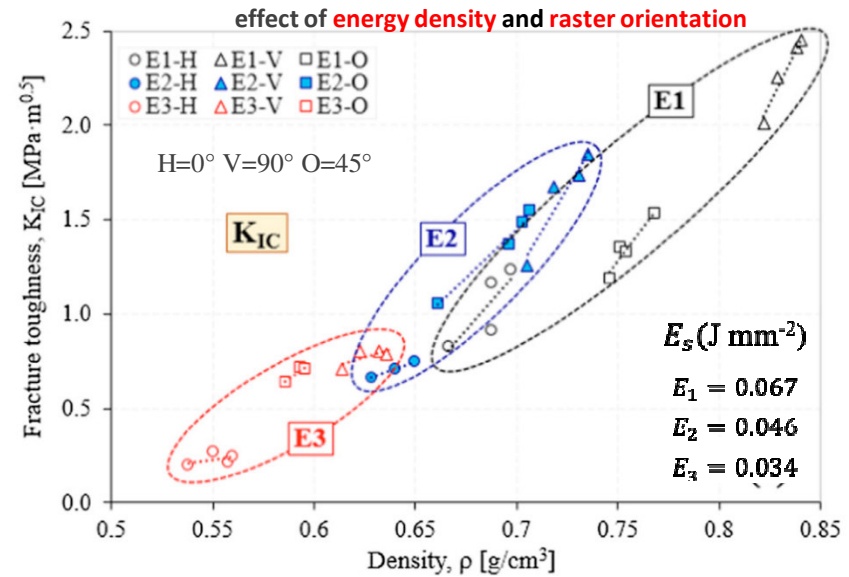
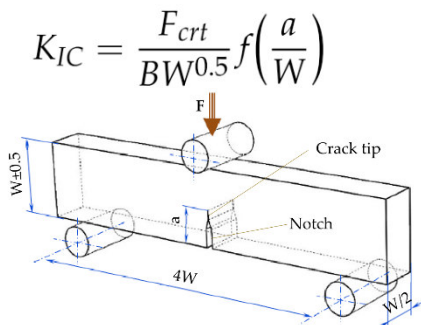
Effect of process parameters on fracture properties



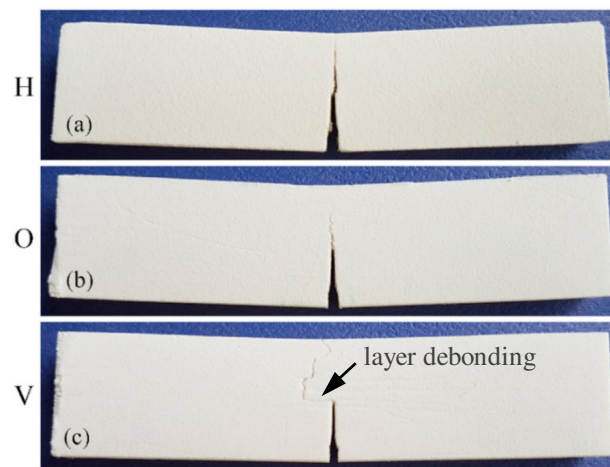
Stoia, D.I. *Polymers*. 2020, 12(3), 640

material: polyamide 12 (PA2200)
 scan spacing $s = 0.25$ mm
 scan velocity $v = 1500-2500$ mm/s

fracture toughness (three-point bending)



Linul, E. *Theor. Appl. Fract. Mech.* 2020, 106, 102497

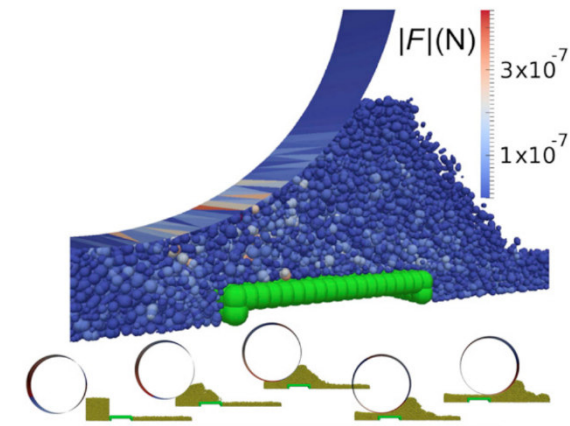
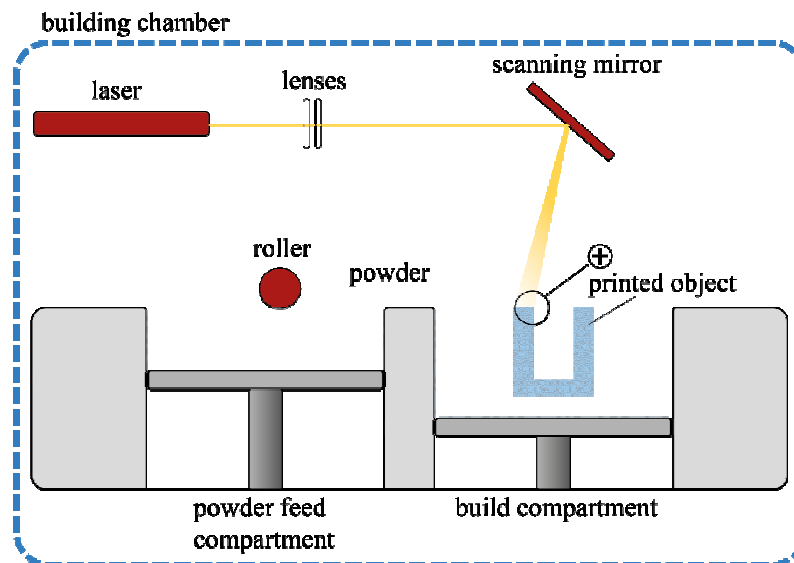


- raster orientation affects the crack path, with layer debonding followed by crack growth in the 90° orientation

2) Simulations

Processes in Selective Laser Sintering:

1. powder spreading ⇒ models of powder recoating
2. energy absorption and heat transfer ⇒ optical and heat transfer models
3. sintering and cooling ⇒ sintering models



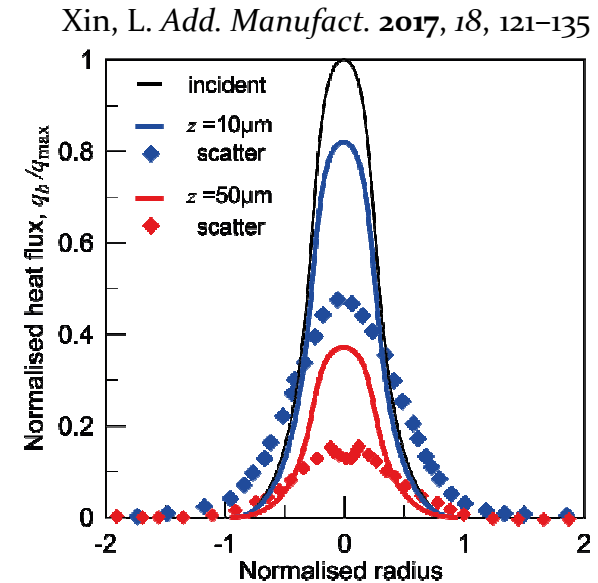
Parteli, E.J.R., et al. *Powder Technol.* 2016, 288, 96–102

- mechanical behaviour of printed parts ⇒ mechanical models

2) Simulations

Optical model

- the optical model describes the laser energy deposition in the powder bed, based on the interaction between the electromagnetic radiation and the **optical properties** of the material
- it should describe the heat flux emitted by a moving light source and the **energy absorption** in depth (according to Lambert-Beer law)
- advanced models account for the effect of powder transparency and **light scattering** through ray-tracing algorithms



surface
heat flux

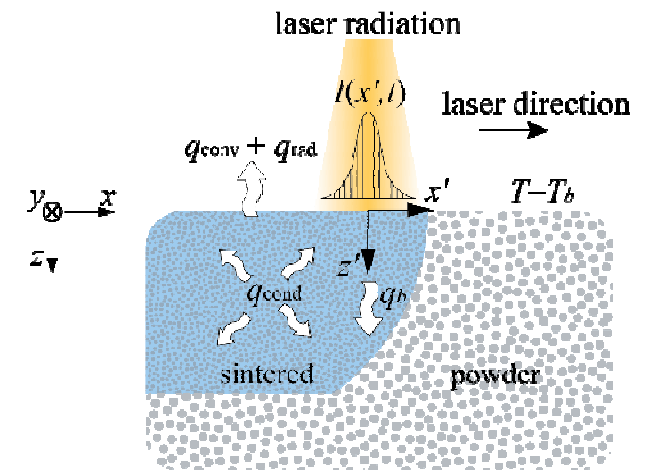
$$I(x, y, t) = I_0 \exp\{-c[(x - vt)^2 + (y - vt)^2]\}$$

↙ maximum light intensity
 ↘ scan velocity

heat flux
density

$$q_b(x, y, z, t) = (1 - R_R) e_R I_0 \exp\{-c[(x - vt)^2 + (y - vt)^2] - e_R z\}$$

↙ attenuation coefficient
 ↘ reflectivity



2) Simulations

Heat transfer model

- heat transfer in the powder bed, including phenomena of conduction, convection and radiation, is described by the equation of **heat conduction**

$$\rho c_p \frac{\partial T}{\partial t} = \nabla \cdot (\lambda \nabla T) + q_g$$

specific heat
thermal conductivity

volumetric heat generation

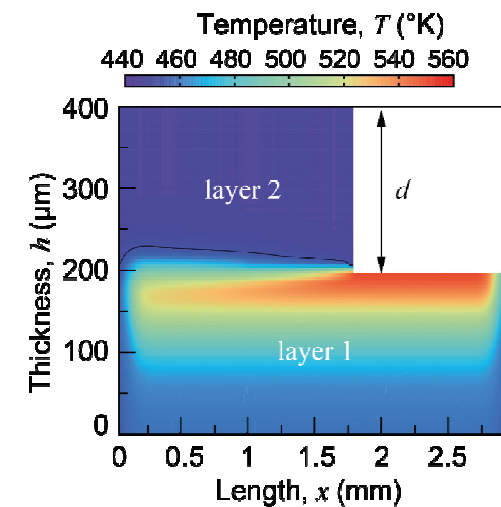
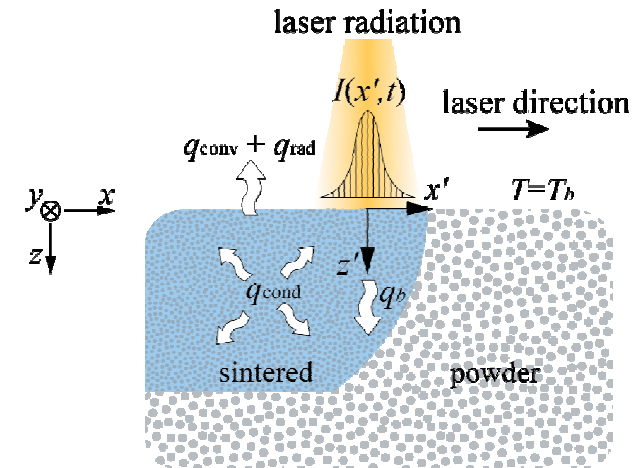
- boundary conditions account for energy losses through **radiation** and **convection** on the powder bed surface
- the **volumetric term** includes the heat flux generated by the laser beam and phase transitions that are associated with **latent heat**

$$q_g = q_b + s_f + s_c$$

laser heat flux
melting

crystallisation

⇒ phase transitions depend on the polymeric material!



2) Simulations

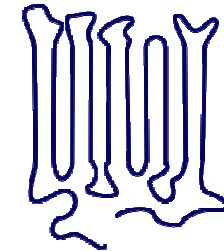
Heat transfer model: **phase transitions**

- a fundamental distinction in the morphology of polymers is between **amorphous** and **semi-crystalline** materials
- the amorphous phase undergoes **glass transition**, upon which the material becomes soft and viscous; the crystalline phase undergoes **melting** and **crystallisation**, which take place at different temperatures



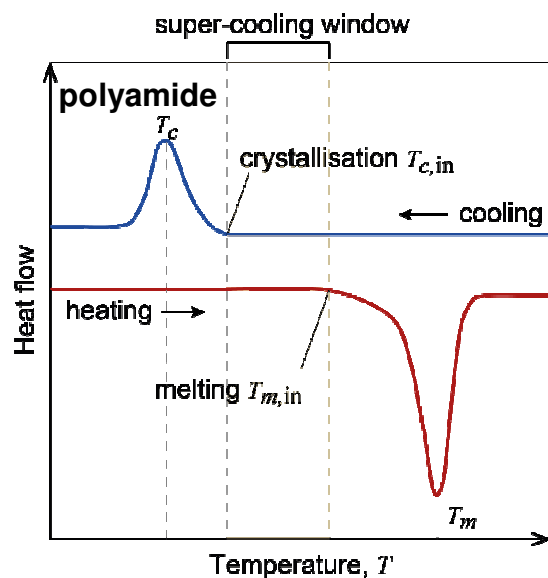
Amorphous

(e.g. polycarbonate, polystyrene)



Semicrystalline

(e.g. polyamide, polyethylene, PEEK)



⇒ optimal processing depends on the polymer's morphology!

- glass transition, as a second-order transformation, is not related to any latent heat

$$\rho c_p \frac{\partial T}{\partial t} = \nabla \cdot (\lambda \nabla T) + q_b$$

heat conduction (amorphous)

- optimal processing of semi-crystalline materials occurs within the so-called **super-cooling window**

$$\rho c_p \frac{\partial T}{\partial t} = \nabla \cdot (\lambda \nabla T) + q_b + s_f + s_c$$

heat conduction (semi-crystalline)



models of crystallisation kinetics

2) Simulations

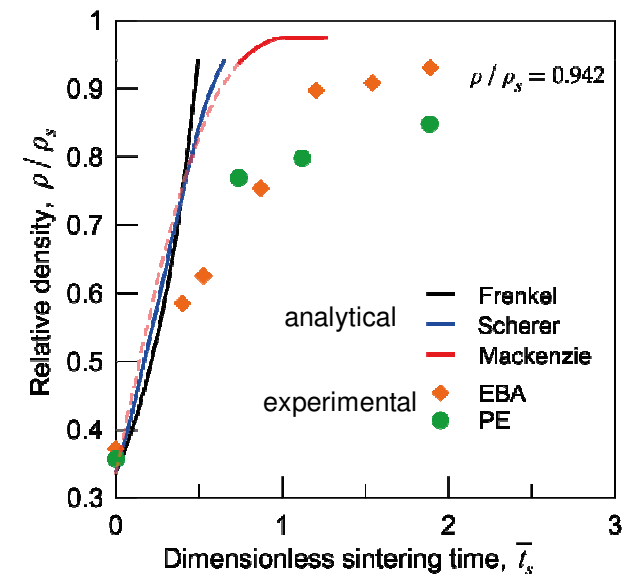
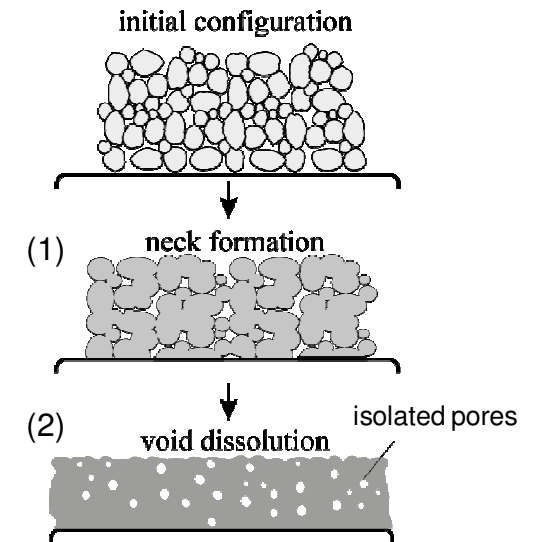
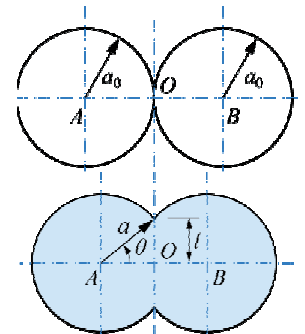
Sintering model

- sintering describes the transformation through which the **polymeric powder is converted into a porous solid**, under the action of surface tension and viscous flow
- densification of the material occurs in **non-isothermal** conditions in two different stages:
 - (1) **powder coalescence**
 - (2) **densification of the molten mass**
- sintering is usually described by empirical relationships:

$$\frac{\partial \rho}{\partial t} = A \exp\left(-\frac{\Delta E}{RT}\right) (\rho_{\infty} - \rho)$$

activation energy ΔE
 theoretical density ρ_{∞}
 universal gas constant R

- recently, particle-based methods implemented **analytical sintering models** in large-scale simulations of SLS (Frenkel, Scherer, etc.)



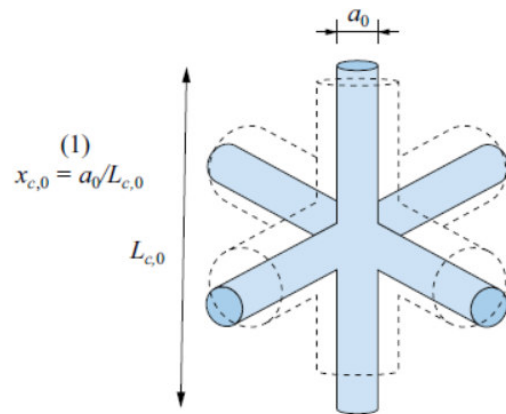
2) Simulations

Sintering models: Frenkel's vs Scherer's

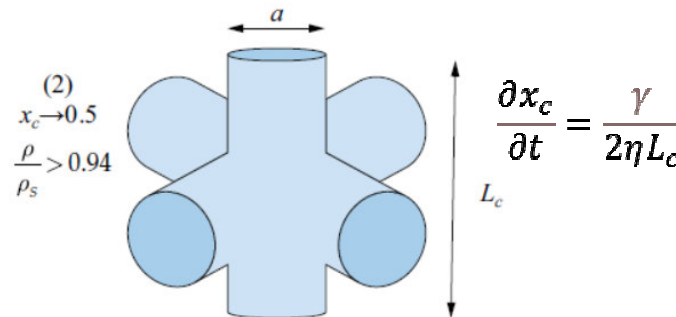
$$\left(\frac{l}{a}\right)^2 = \frac{\gamma}{\eta a_0} t$$

surface tension γ

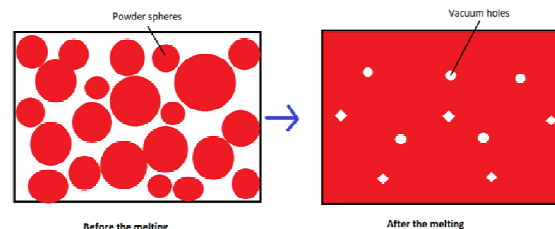
viscosity: $\eta = \eta_0 \exp\left(\frac{\Delta E}{RT}\right)$



Scherer GW I. *Theory. J Am Ceram Soc* 1977 60(5-6):236-239

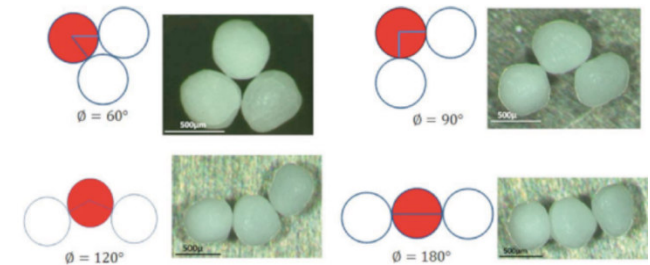
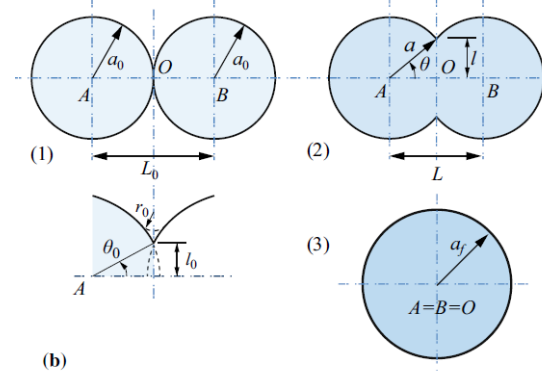


Sintering models: bubble dissolution (e.g. see Mackenzie's model)



Mackenzie JK, Shuttleworth R (1949) *Proc Phys Soc B* 62(12):833-852.


Frenkel JJ *J Phys (USSR)* 1945 9:385-391



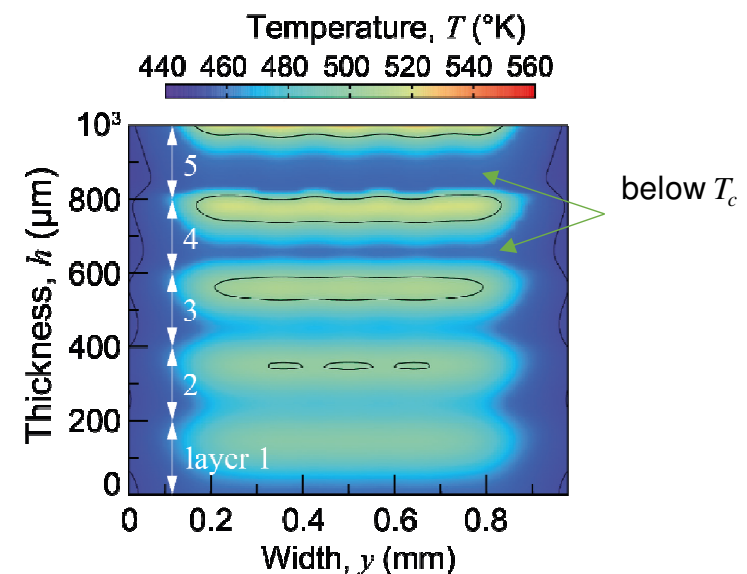
2) Simulations

Models of the mechanical behaviour

- the fundamental stage in which the material develops its mechanical properties is the **cooling phase**. However, due to the multi-layered construction, temperature evolves non-uniformly during printing, resulting in residual stresses and inhomogeneous crystallisation


 the **mechanical problem is coupled with the thermal and sintering problems** because the material properties evolve during printing

- the main aspects to be addressed are:
 - role of **part porosity**
 - role of the **crystalline phase**
 - time- and temperature-dependence
- most existing approaches consider the mechanical problem separated from the printing process!



Mokrane, A. *CR Mecanique* 2018, 346(11), 1087-1103

material: polyamide 12 (PA2200)

melting temperature $T_m = 461^\circ\text{K}$

crystallisation temperature $T_c = 454^\circ\text{K}$

2) Simulations

Models of the mechanical behaviour: **role of crystallisation**

- the mechanical behaviour of semi-crystalline polymers subjected to tensile loading is governed by rupture and reorientation in the crystalline regions, followed by deformation in the amorphous region
- the resulting time- and temperature-dependent response can be modelled by **generalised viscoelastic models** and **time-crystallisation-temperature superposition**

relaxation modulus

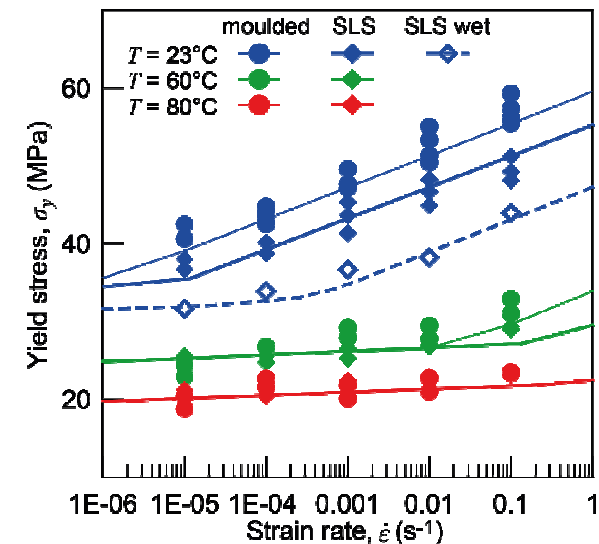
$$\mu(t, \alpha, T) = \mu_{\infty} + \sum_i \mu_i(\alpha_{\text{ref}}, T) \exp\left(-\frac{t}{A_C(\alpha, T)\tau_i}\right)$$

degree of crystallinity \downarrow μ_{∞}
 shift function \swarrow $A_C(\alpha, T)$ \searrow relaxation time τ_i

- the yield stress of semi-crystalline materials can be expressed with a modified form of Ree-Eyring's **activated flow theory**

$$\sigma_y(\dot{\epsilon}, T) = \frac{k_B T}{V_1^*} \sinh^{-1} \left[\frac{\dot{\epsilon}}{\dot{\epsilon}_{0,1}} \exp\left(\frac{\Delta E_1}{RT}\right) \right] + \frac{k_B T}{V_2^*} \sinh^{-1} \left[\frac{\dot{\epsilon}}{\dot{\epsilon}_{0,2}} \exp\left(\frac{\Delta E_2}{RT}\right) \right]$$

\downarrow deformation of crystalline phase \downarrow deformation of amorphous phase



2) Simulations

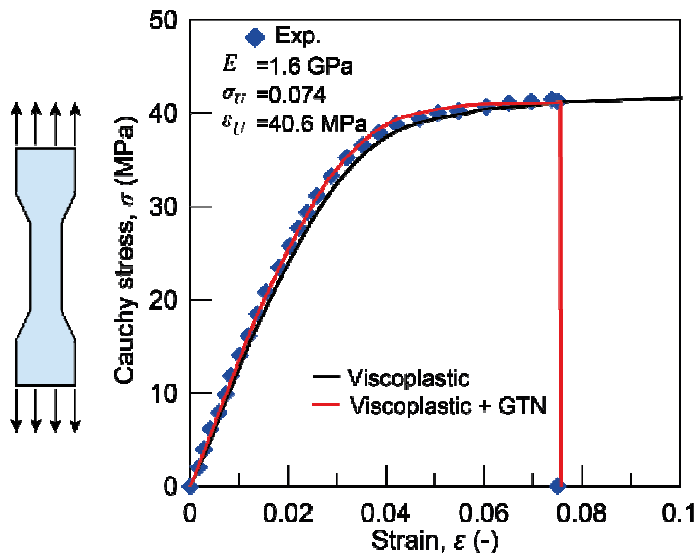
Models of the mechanical behaviour: **role of porosity**

- depending on the microstructure, failure of porous materials under tensile conditions occurs with either brittle or ductile mechanisms
- ductile failure can be modelled with a **mechanism of void growth**, according to Gurson-Tvergaard-Needleman (GTN) damage model

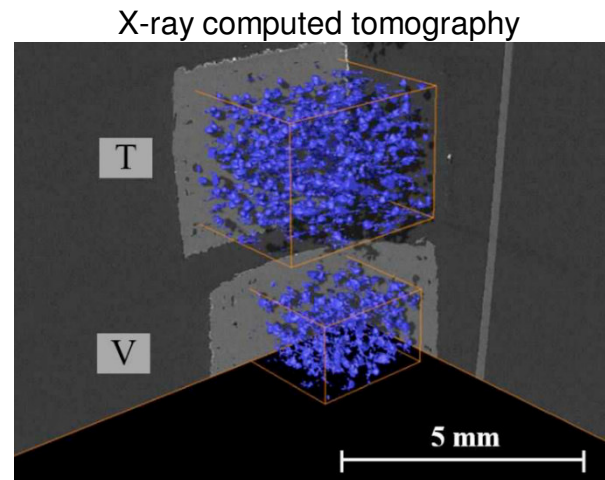
GTN yield function

$$\psi(\sigma, \sigma_y, \phi) = \left(\frac{\sigma_{VM}}{\sigma_y}\right)^2 + 2q_1\phi^* \cosh\left(\frac{3}{2}q_2\frac{\sigma_h}{\sigma_y}\right) - (q_3\phi^{*2} + 1) = 0$$

yield stress modified void fraction $q_1, q_2, q_3 = \text{temperature-dependent parameters}$



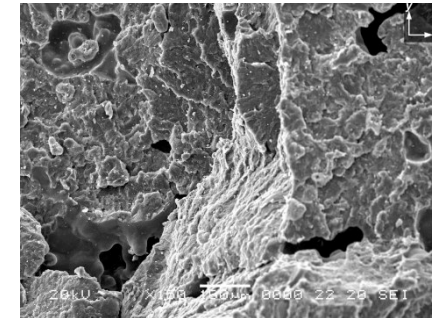
Schob, D. *Eng. Fract. Mech.* 2020, 229, 106841



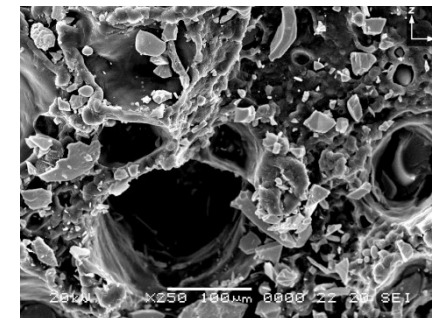
pore distribution in virgin (V) and under tension (T)

Schob, D. *Arch. Mech.* 2019, 71(4-5), 507-526

brittle fracture



ductile fracture



Leigh, K. 2012. 23rd Annu. Int. Solid Free. Fabr. Symp. 574-605

material: polyamide 12 (PA2200)

6. PHOTOPOLYMERIZATION (SLA, DLP): EXP/SIMULATIONS

Photopolymerization is a technique that uses light (visible or ultraviolet, UV) to initiate and propagate a polymerization reaction to form a linear or crosslinked polymer structure.



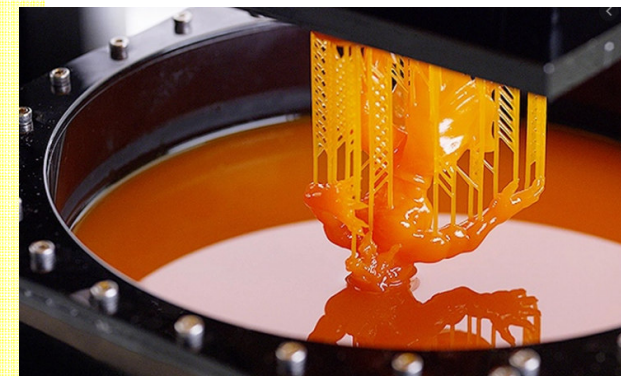
Printable polymers:

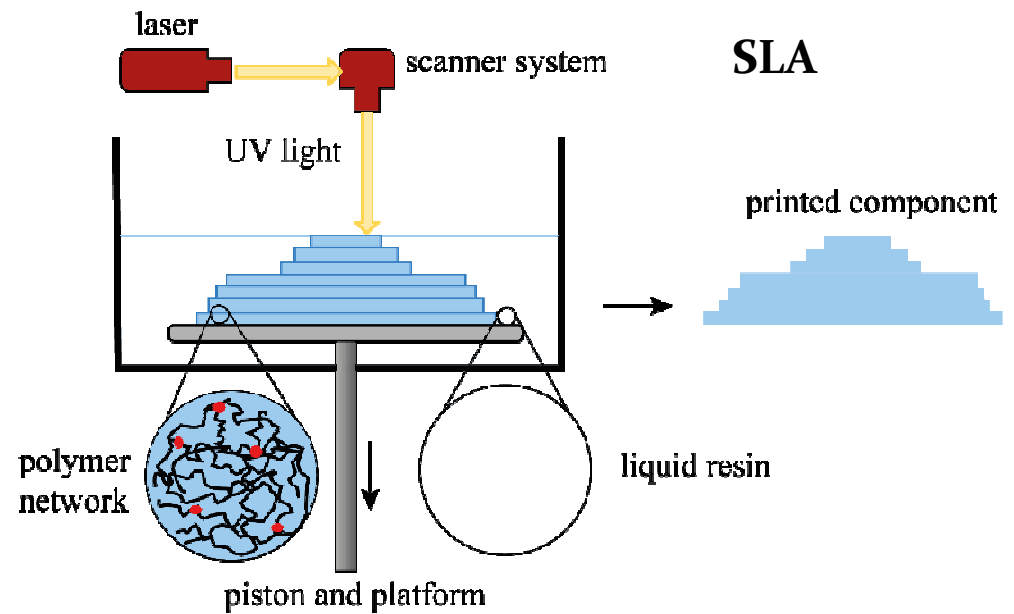
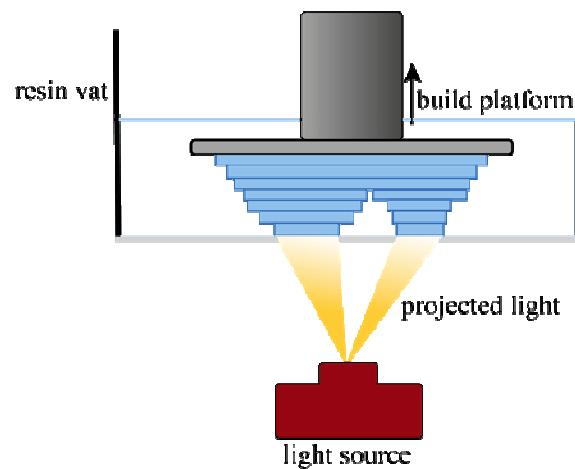
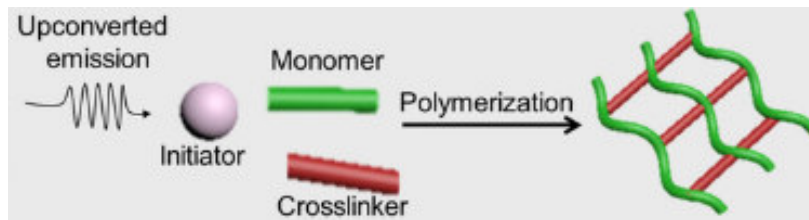
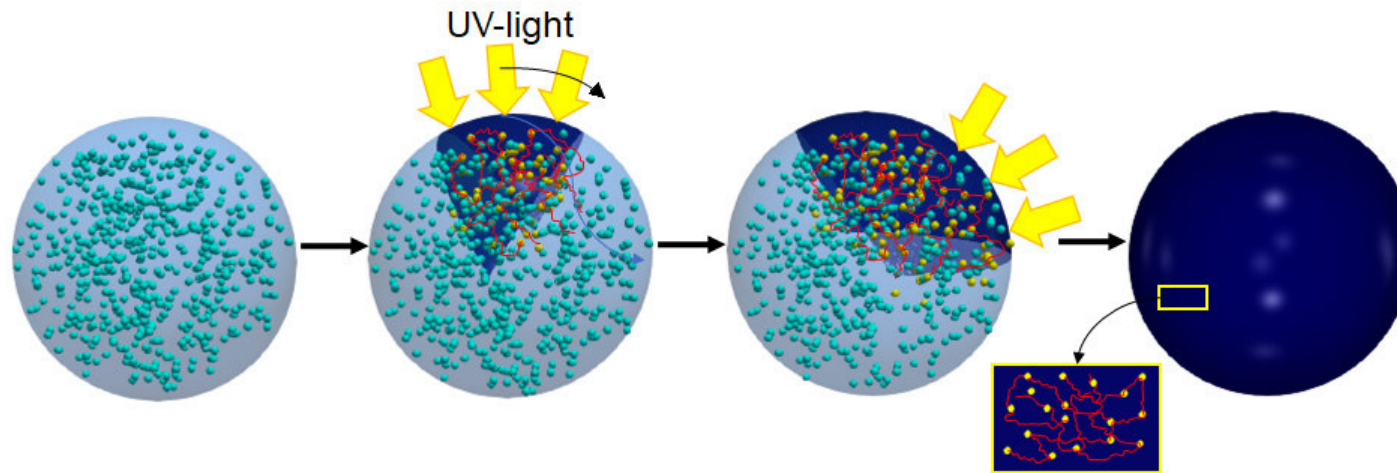
- all the types of photopolymer resins, such as polyester acrylate (PEA), epoxy acrylates (EA), urethane acrylates (UA).

Many commercial resins for SL are available

For instance, some commercial SL-resins are based on:

- PEGDA: Poly(ethylene glycol) diacrylate monomer
- PTMC: Poly(trimethylene carbonate) monomer
- HDDA: hexamethylene diacrylated monomer



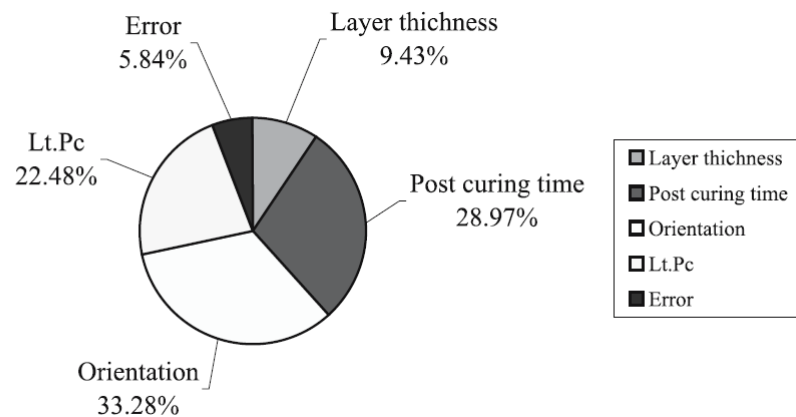


Digital Light Processing (DLP)

1) Experiments

Parameter	Range	Level 1	Level 2	Level 3
<i>Lt</i>	0.1–0.15 mm	0.1	0.125	0.15
<i>Pc</i>	60–120 min	60	90	120
<i>O</i>	HX, VX, HY, VY	HX	VX	HY

- Layer thickness
- Post-curing time
- Building orientation



Influence of the selected input AM parameters on the mechanical properties

Experiment number	Level		
	Layer thickness (<i>Lt</i>)	Post-curing time (<i>Pc</i>)	Orientation (<i>O</i>)
1	1	1	1
2	1	2	2
3	1	3	3
4	2	1	1
5	2	2	2
6	2	3	3
7	3	1	2
8	3	2	3
9	3	3	1
10	1	1	3
11	1	2	1
12	1	3	2
13	2	1	2
14	2	2	3
15	2	3	1
16	3	1	3
17	3	2	1
18	3	3	2

ANOVA statistical analysis

1) Experiments

Figure 1 Tensile test specimen **YS and UTS**

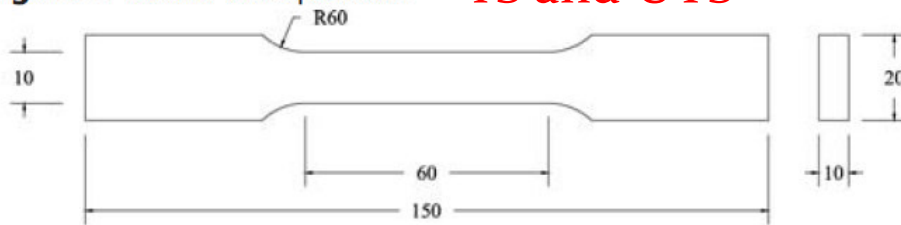
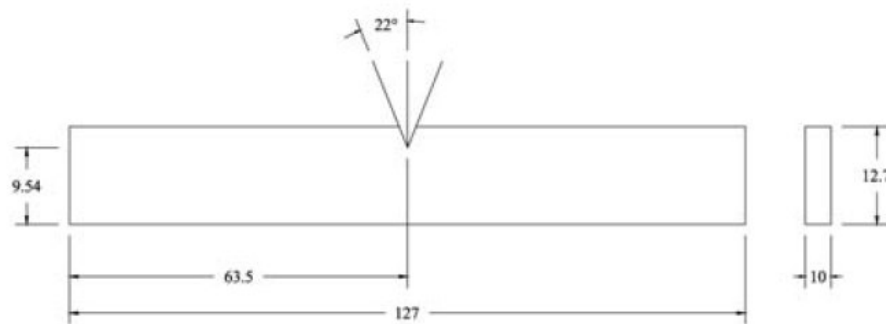


Figure 2 Impact test specimen **E and IS**



Chockalingam, K. *The International Journal of Advanced Manufacturing Technology* 2006, 29(1-2), 79-88

Effect of layer thickness on:

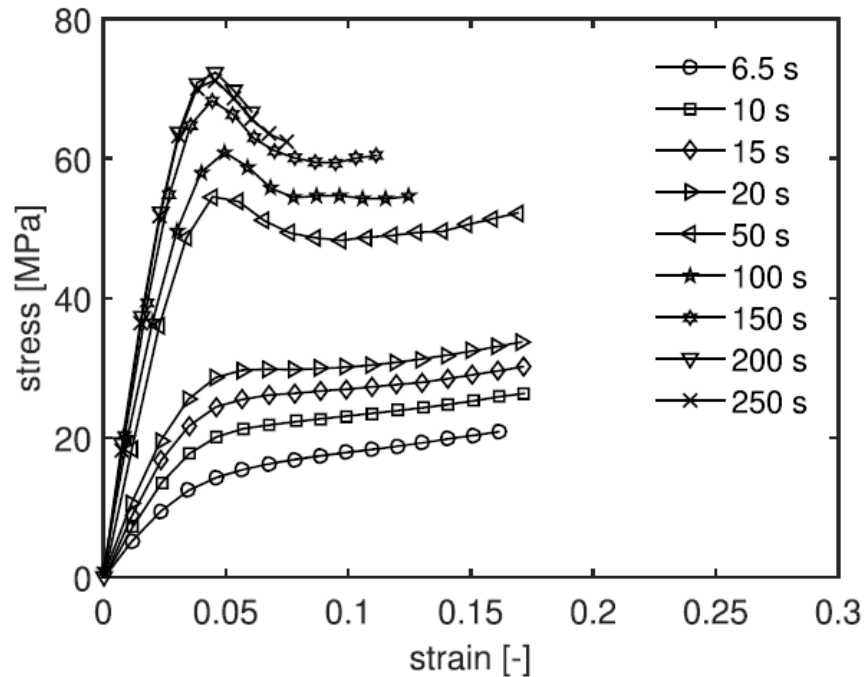
- Yield strength (YS)
- Ultimate tensile strength (UTS)
- Impact toughness (E)
- $IS = E/W$ (W=width of the specimen= 10 mm)

Layer thickness

Test Number	Lt (μm)	Average values of ten specimens			
		Average YS (N/mm^2)	Average UTS (N/mm^2)	Average E in (J)	Average IS in (J/m)
1	100	68.24	70.05	0.2926	29.25
2	125	60.76	68.50	0.2539	25.39
3	150	49.92	56.00	0.2125	21.25

Optimum value
(best mechanical properties)

1) Experiments

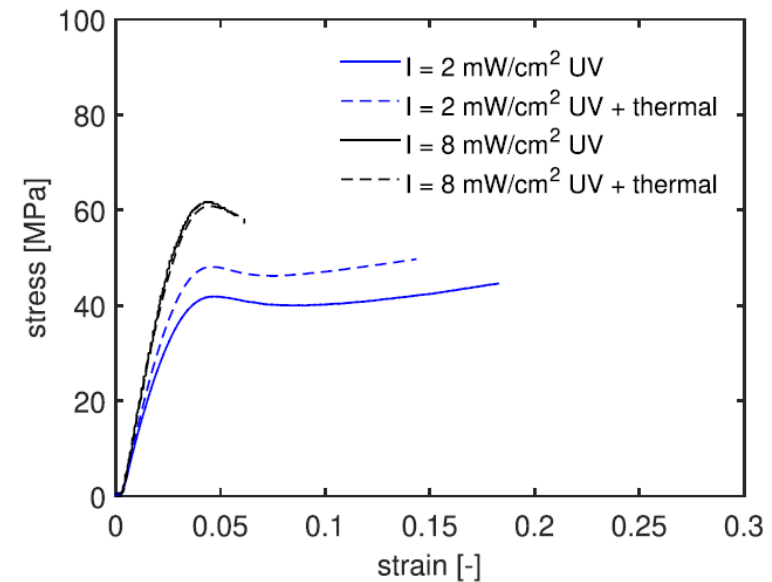


Stress-strain curves for different curing time at a fixed light intensity of 8 mW/cm^2

On the material printed with the highest light intensity, no effect of post-curing time

Why...? Theoretical models are needed to explain this aspect

- UV-light intensity
- Curing time
- Post-curing time



Effect of post-curing time

1) Experiments

- Building orientation

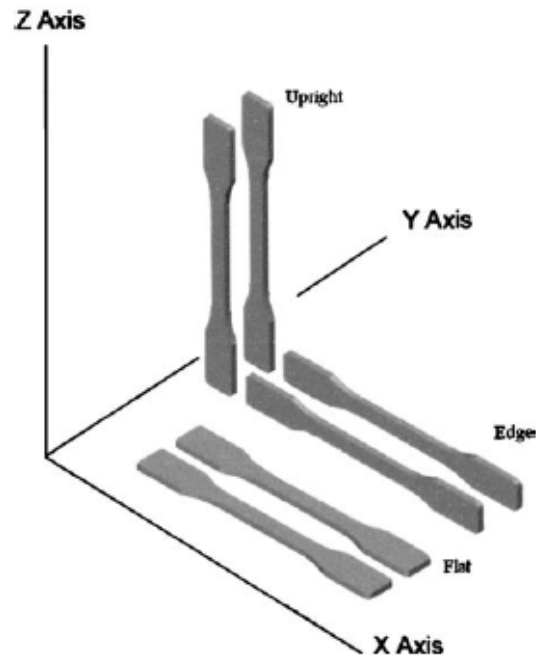


TABLE I Isotropy/anisotropy test results for SL 7560

Mechanical properties	Build orientation			Max. % variation
	Flat	Edge	Upright	
Max. tensile strength (Mpa)	54.9	56.4	53.7	4.8
Young's modulus (GPa)	2.6	2.7	2.7	3.7
Flexural strength (MPa)	92.5	96.3	95.3	3.9
Flexural modulus (GPa)	2.1	2.2	2.1	4.5
Impact strength (kJ/m ²)	2.5	2.4	2.4	4.0

So, a photopolymerized-component is **ISOTROPIC** or **ANISTROPIC**?

The problem is still under investigation...

2) Simulations

Polymer **chain length and crosslink density** evolve according to the **degree of cure** evolution



The mechanics of a polymer depends on the amount of the degree of cure

$$\rho(\mathbf{x}, t) = 1 - \frac{C_M(\mathbf{x}, t)}{C_M(\mathbf{x}, t = 0)}$$

KINETIC MODELS OF PHOTOPOLYMERIZATION (recently developed)

They aim at calculating in different ways the degree of cure, in order to relate it with the mechanics of a printed component

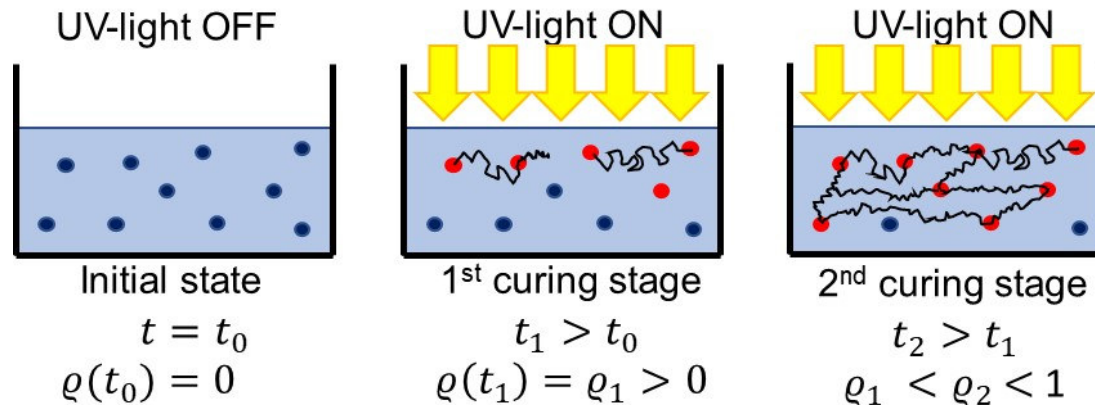
- **Phenomenological**

The degree of cure evolution is modelled by means of **only one** differential equation (for as concerning the kinetic model of photopolymerization)

- **Mechanistic**

The degree of cure is obtained by solving a **system of differential equations** taking into account the concentration evolution of the species involved in the problem

2) Simulations



- Photo-initiators in the inactive state
- Free radicals

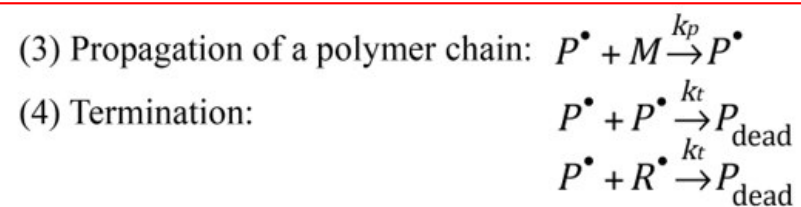
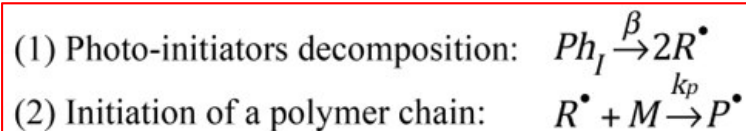
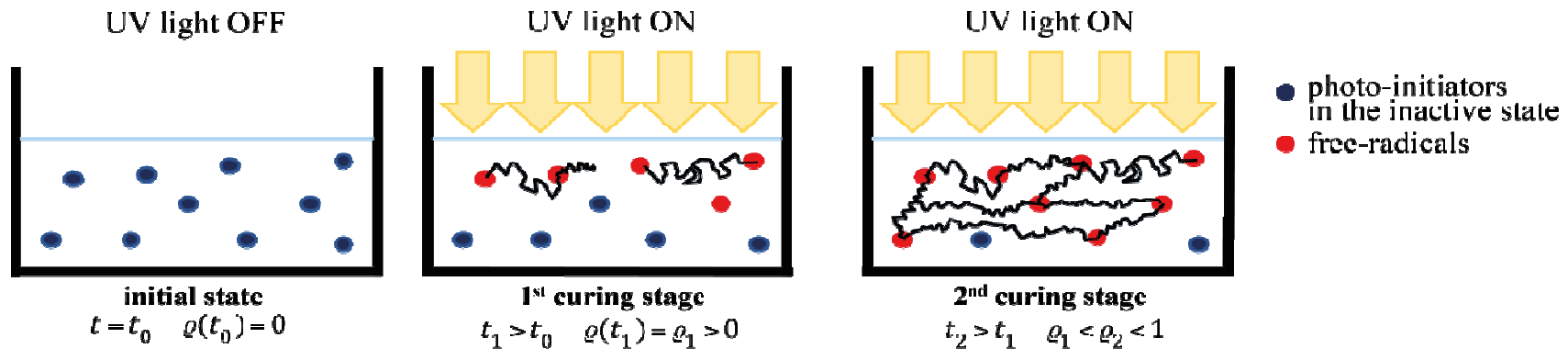
The mechanical behavior of a component printed with Photopolymerization depends by **the degree of cure (ϱ)** achieved during the printing process itself, which is affected by the process parameters adopted

What physically represents the **degree of cure**? It is an **internal variable** (typically expresses as a percentage) used for quantifying:

- Quantity of monomer **conversion**
- Quantity of $C = C$ bonds (liquid state) converted in $C - C$ bonds (polymeric state) as typically happens in acrylates photopolymerization

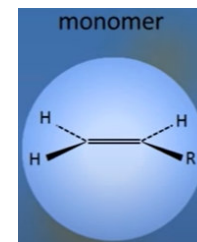
2) Simulations

Simulation of the photopolymerization process (SLA)

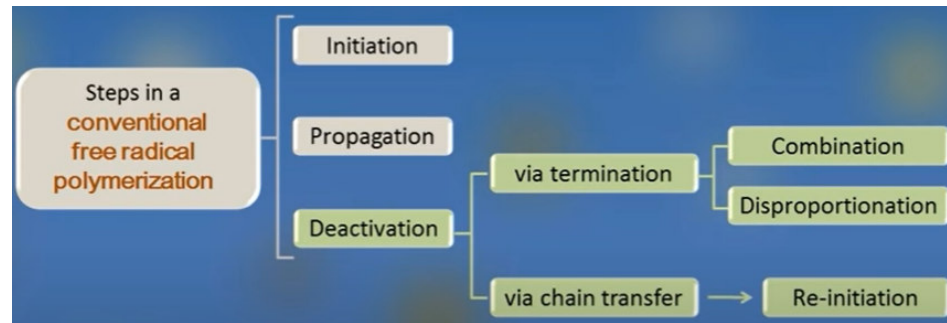


We need to describe a multiphysics problem involving:

- 1) Light propagation (Beer-Lambert law)
- 2) Kinetics of chemical species evolution
- 3) Chain density (shear modulus)



2) Simulations



- Photo-initiators rate \dot{C}_I
(consumed to provide activation of free radicals)

$$\dot{C}_I(x, t) = -\beta I(x, t) C_I(x, t)$$

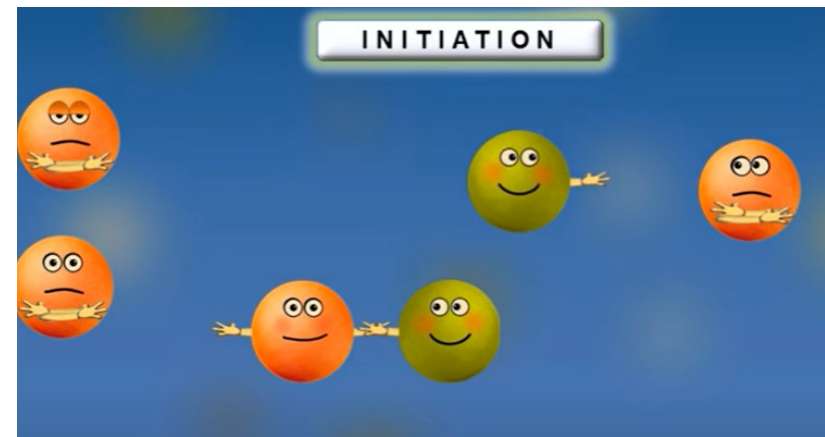
β : photodecomposition rate $\dot{\blacksquare} = \frac{\partial \blacksquare}{\partial t}$

- Evolution of the free radicals C_R

$$\dot{C}_R(x, t) = +m \dot{C}_I(x, t) - m k_t(x, t) [C_R(x, t)]^2$$

+

-



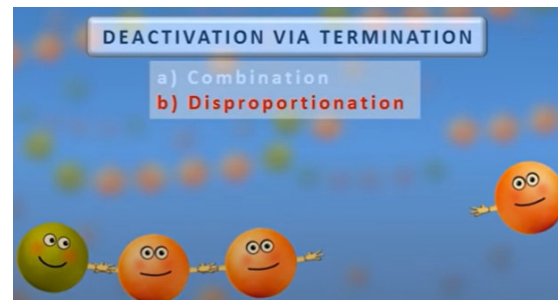
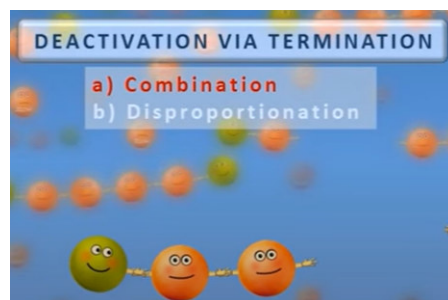
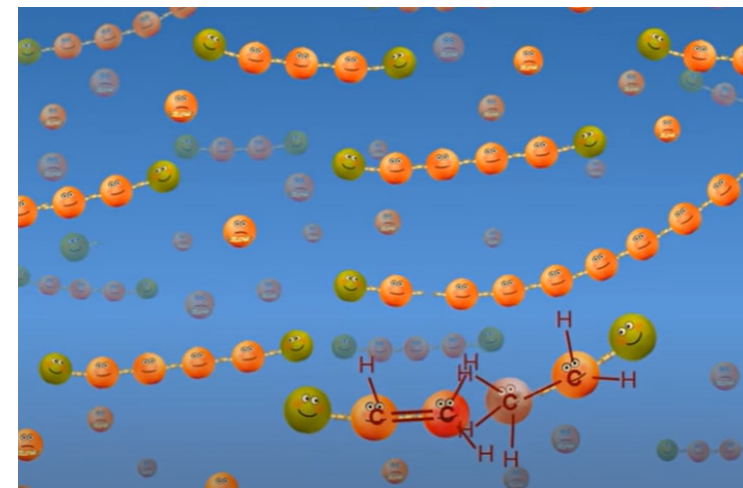
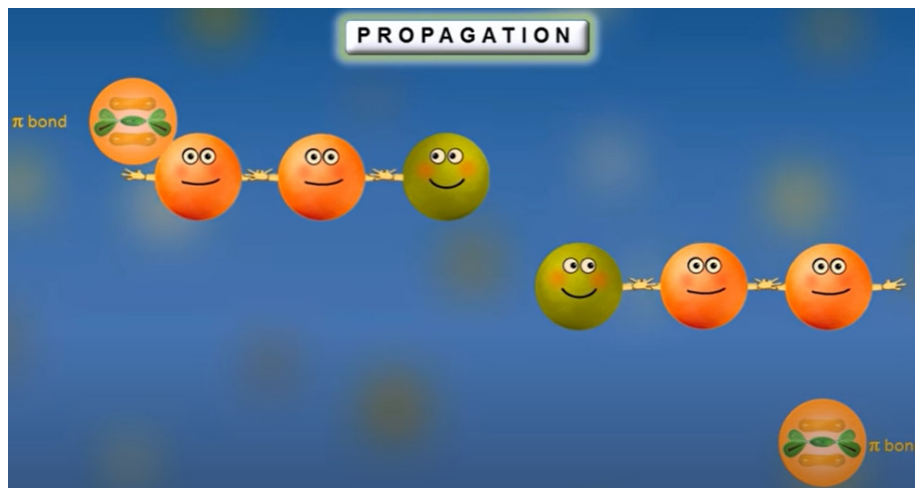
m : number of radicals generated in the photodecomposition (e.g. $m = 2$)

k_t : termination rate

- As the combination with radicals proceeds, the monomers C_M in the solution are gradually consumed, allowing the polymer chains initiation and propagation to occur:

$$\dot{C}_M(x, t) = -k_p(x, t)C_M(x, t)C_R(x, t)$$

k_p : propagation rate constant



2) Simulations

1) Light propagation (Beer-Lambert law):

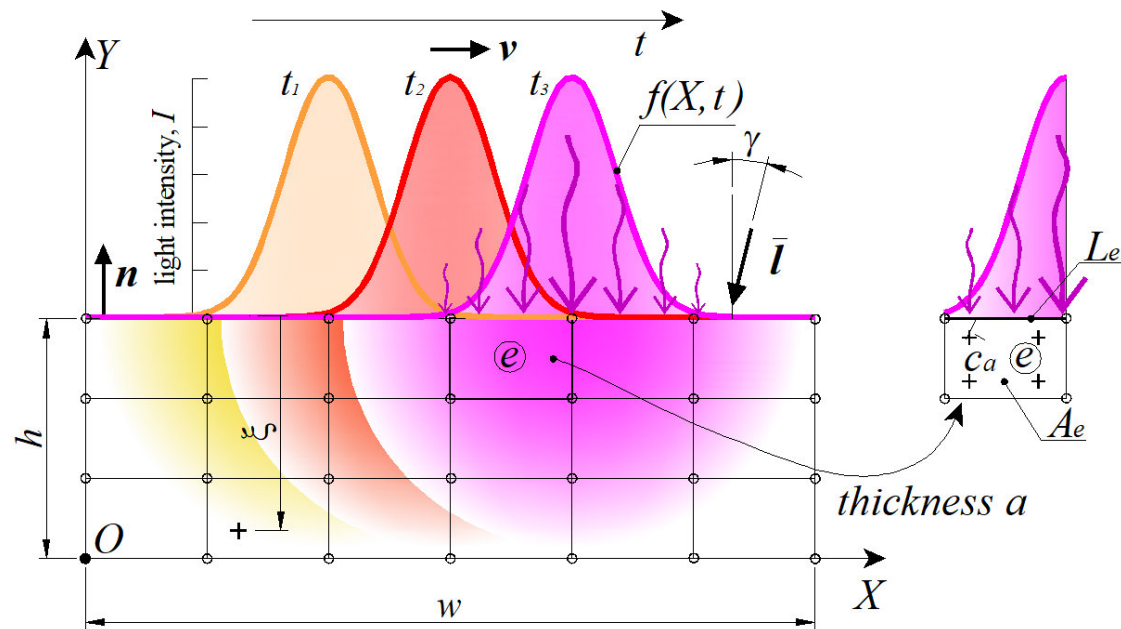
Governing field equations

$$I(\mathbf{X}, t) \cdot \nabla_{\mathbf{X}} I(\mathbf{X}, t) = -A(\mathbf{X}, t) I(\mathbf{X}, t)$$

for $\mathbf{X} \in \Omega_0$

Boundary conditions:

$$I(\mathbf{X}, t) = I_0(\mathbf{X}, t) \quad \text{for } \mathbf{X} \in \partial\Omega_0$$



The solution for the light intensity field in time can be obtained by **FEM implementation** of the above equations

depletion matrix

$$[\mathbf{E}_e(t) + \mathbf{A}_e(t)] \tilde{\mathbf{I}}_e(t) = \mathbf{P}_e(t)$$

light gradient matrix

$\tilde{\mathbf{I}}_e(t)$ Nodal values of the light intensity

$\mathbf{P}_e(t)$ nodal values of the distribution of the incoming light intensity

FEM coding

$$[\mathbf{E}_e(t) + \mathbf{A}_e(t)] \tilde{\mathbf{I}}_e(t) = \mathbf{P}_e(t)$$

or, equivalently:

$$\left[\int_{V_e} [B]^T \bar{\mathbf{l}}(t) [N] dV - \int_{V_e} [N]^T A(\mathbf{X}, t) [N] dV + Qh_e \int_{V_e} [B]^T [B] dV \right] \tilde{\mathbf{I}}_e(t) \\ = \int_{S_e} [N]^T f_n(\mathbf{X}, t) dS$$

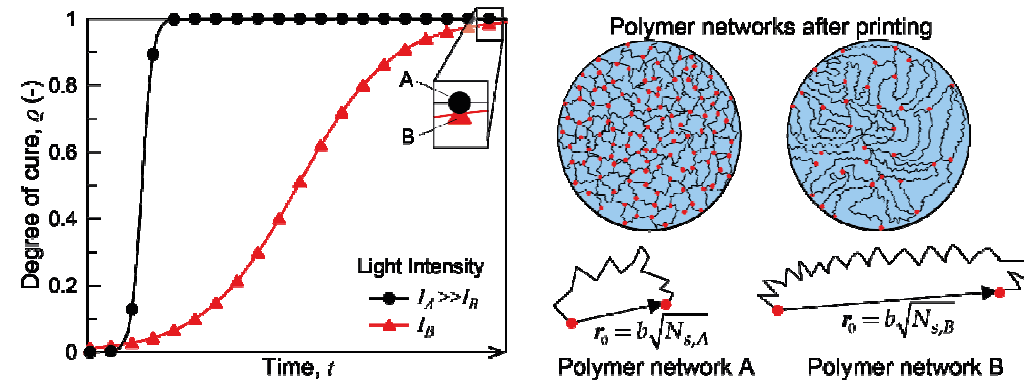
Explicitly, the above matrices are obtained as follows:

$$\mathbf{E}_e(t) = \int_{V_e} [B]^T \bar{\mathbf{l}}(t) [N] dV, \quad \mathbf{A}_e(t) = - \int_{V_e} [N]^T A(\mathbf{X}, t) [N] dV,$$

$$\mathbf{D}_e = Qh_e \int_{V_e} [B]^T [B] dV \quad \text{and} \quad \mathbf{P}_e(t) = \int_{S_e} [N]^T f_n(\mathbf{X}, t) dS$$

2) Simulations

- Evaluation of the degree of cure



$$\dot{\rho}(\mathbf{x}, t) = \frac{k_p(\mathbf{x}, t)C_M(\mathbf{x}, t)C_R(\mathbf{x}, t)}{C_{M0}} \quad \rightarrow \quad \rho(\mathbf{X}, t) = 1 - \frac{C_M(\mathbf{X}, t)}{C_M(\mathbf{X}, t=0)}$$

- Evaluation of chain concentration

Shear modulus
(Fully cured polymer)

$$c_a(\mathbf{X}, t) = \frac{\mu(\mathbf{X}, t)}{k_B T} = \frac{\bar{\mu}}{k_B T} \cdot \exp[\alpha(\rho(\mathbf{X}, t) - 1)]$$

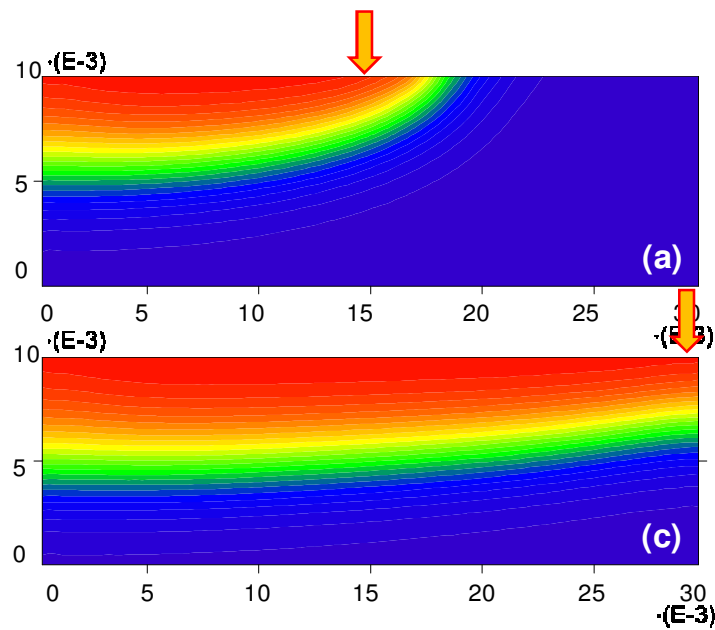
$$c_a(\mathbf{X}, t_c) = c_{a0}(\mathbf{X}) + \int_0^{t_c} \dot{c}_a(\mathbf{X}, t) dt$$

$$\begin{aligned} \mu(\mathbf{X}, t) &= G(\mathbf{X}, t) = \\ &= c_a(\mathbf{X}, t) k_B T \end{aligned}$$

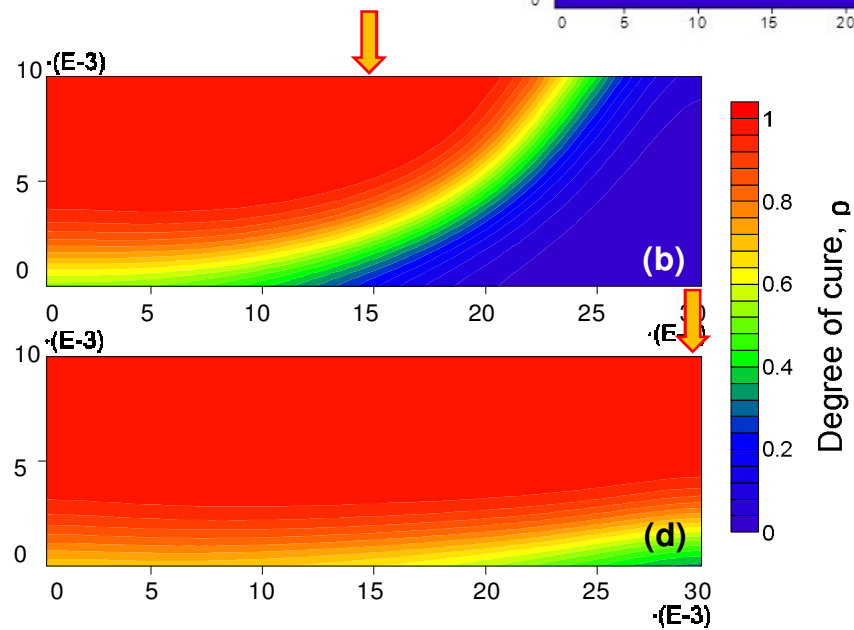
Polymer shear modulus distribution at time t

2) Simulations

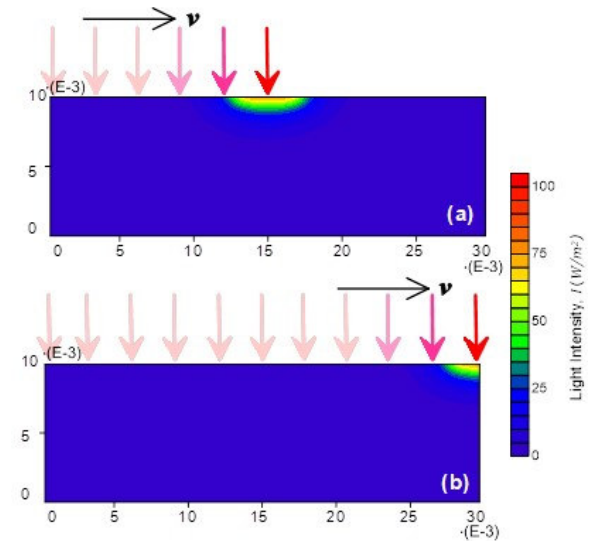
Distribution of the degree of cure



$$v = 10^{-3} \text{ m/s}$$

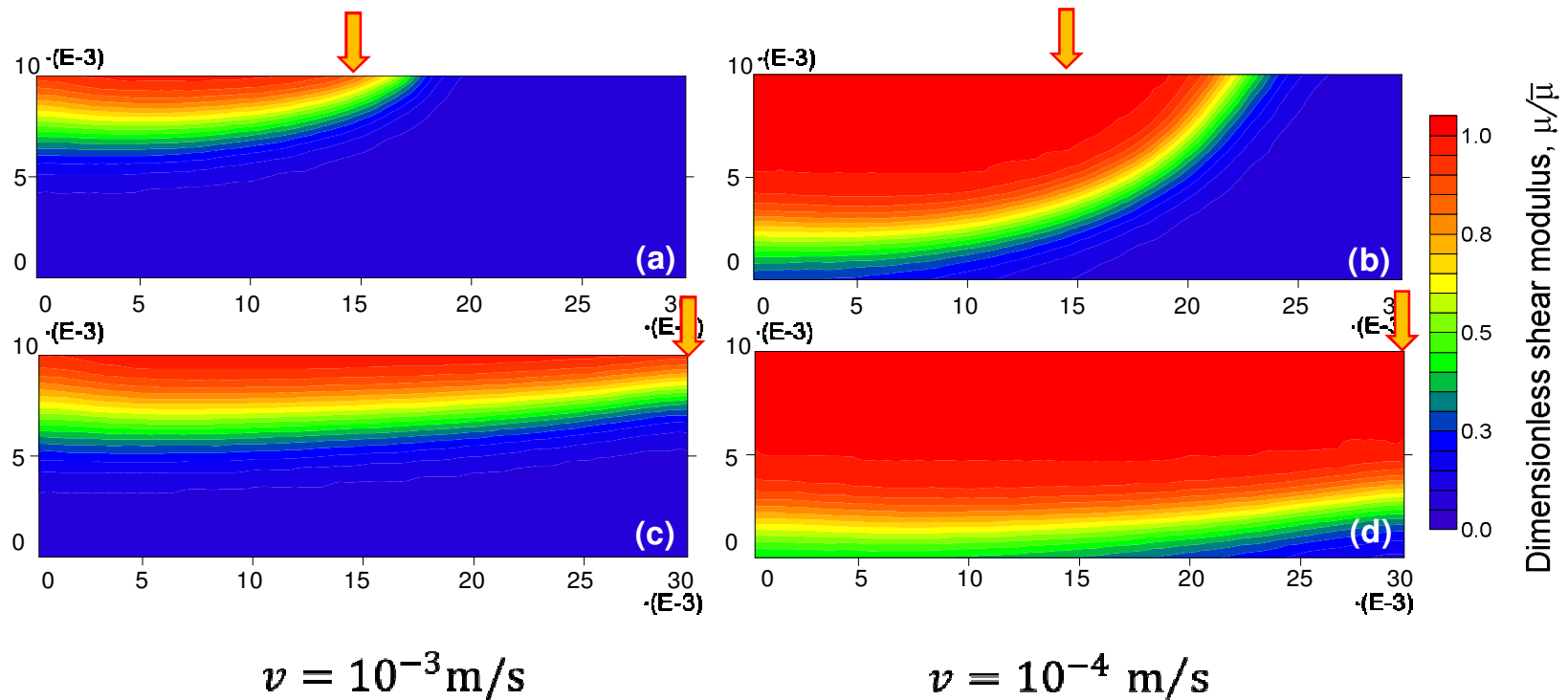


$$v = 10^{-4} \text{ m/s}$$



2) Simulations

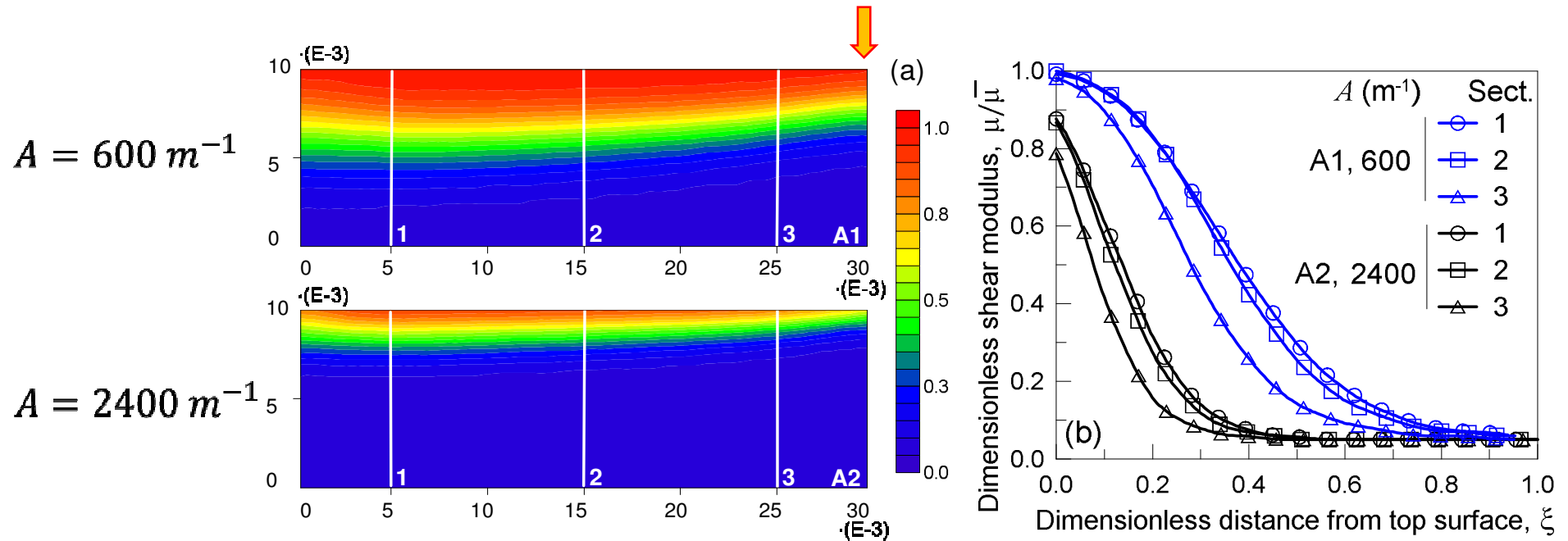
Final distribution of the shear modulus



2) Simulations

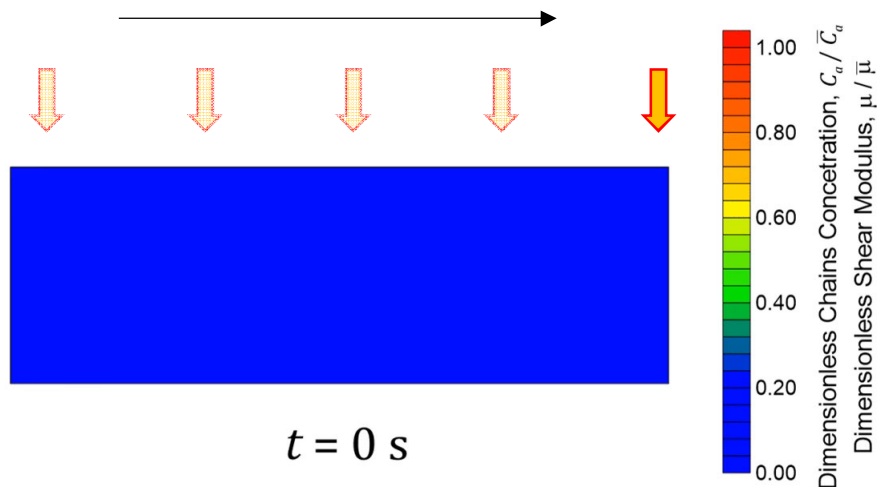
Final distribution of the shear modulus

$$I_0 = 150 \text{ W/m}^2$$



2) Simulations

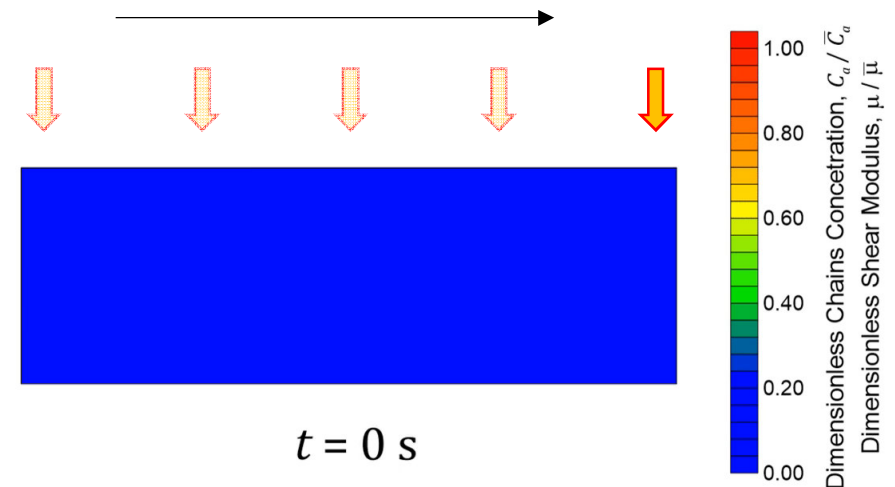
Real-time evolution of the shear modulus in SLA


 $t = 0$ s

$$v = 10^{-3} \text{ m/s}$$

$$I_0 = 150 \text{ W/m}^2$$

$$A = 600 \text{ m}^{-1}$$

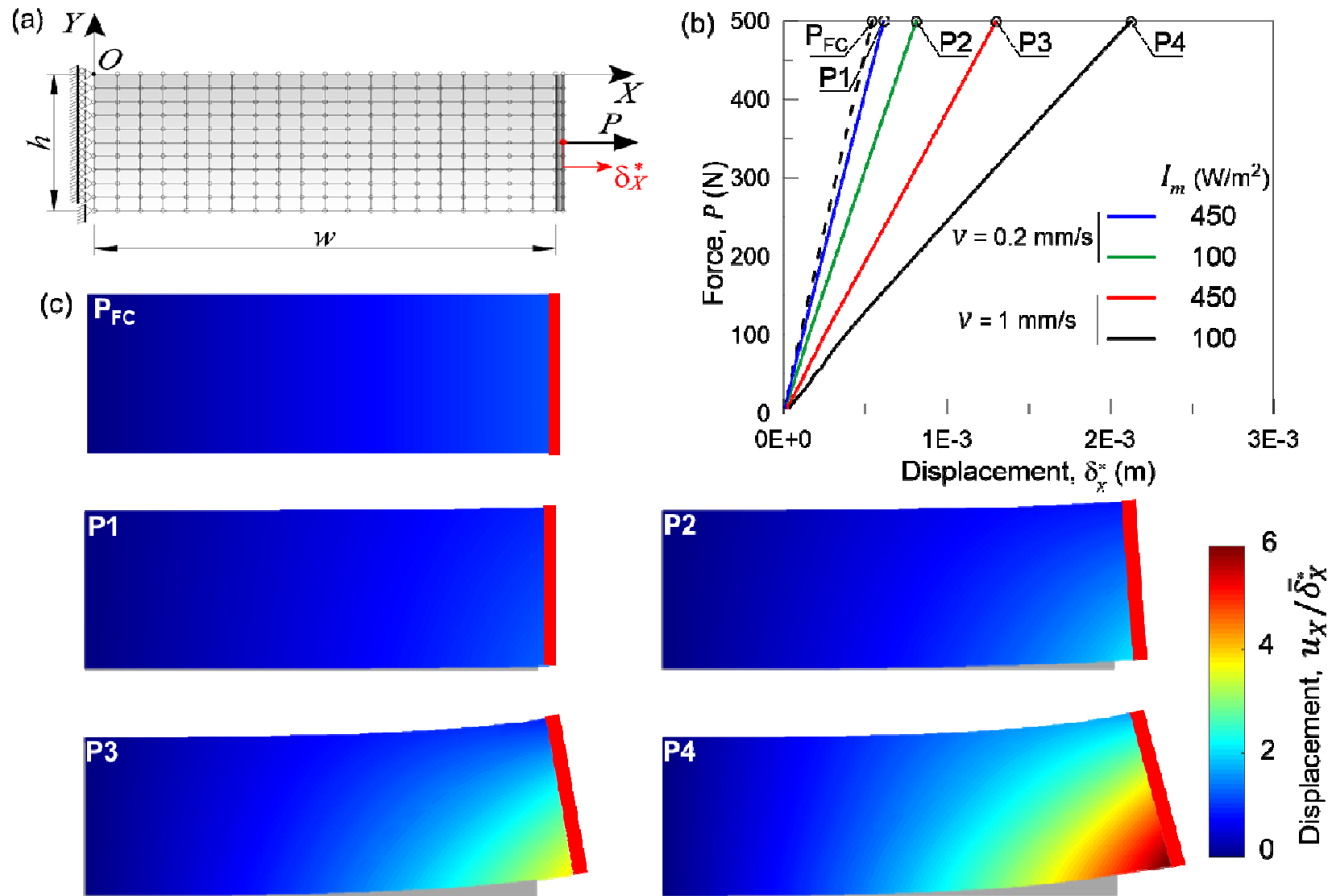

 $t = 0$ s

$$v = 10^{-4} \text{ m/s}$$

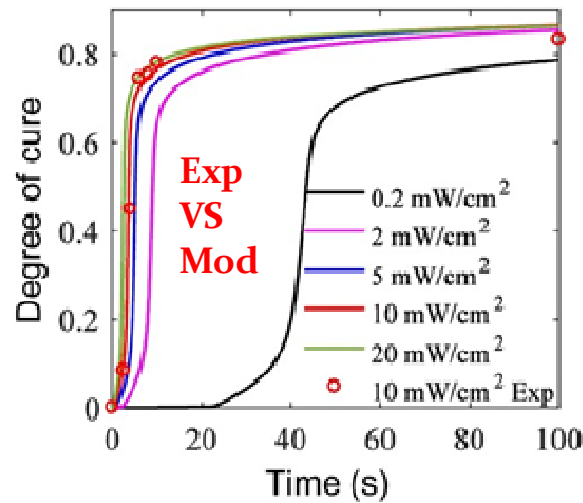
$$I_0 = 150 \text{ W/m}^2$$

$$A = 600 \text{ m}^{-1}$$

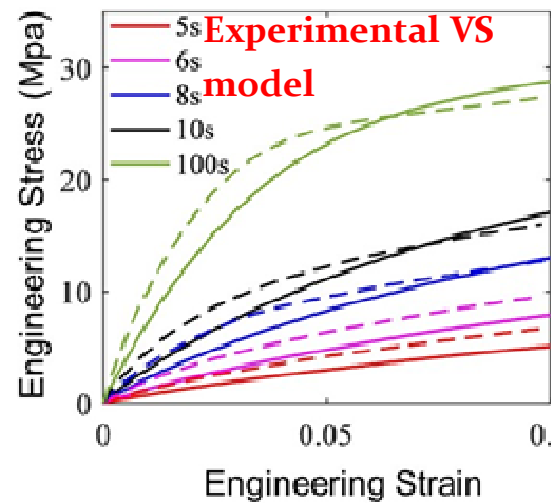
Influence of the AM setup on the mechanical response of a tensile bar



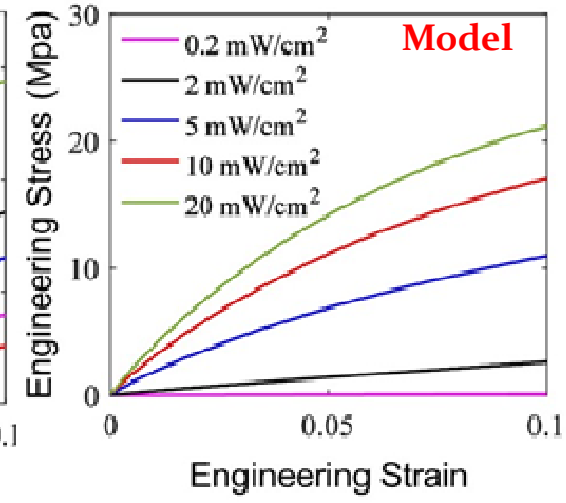
2) Simulations (literature models)



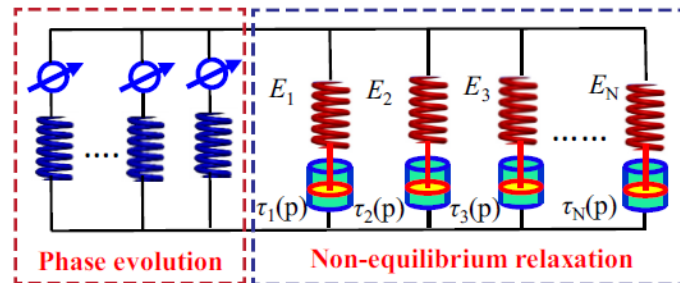
(a)



(b)



(c)



$$\sigma_{eq} = J^{-1}(\mu_{eq} J^{-\frac{2}{3}} FF^T + k_{eq} \ln J I)$$

Constitutive parameters
(Functions of the degree of cure)

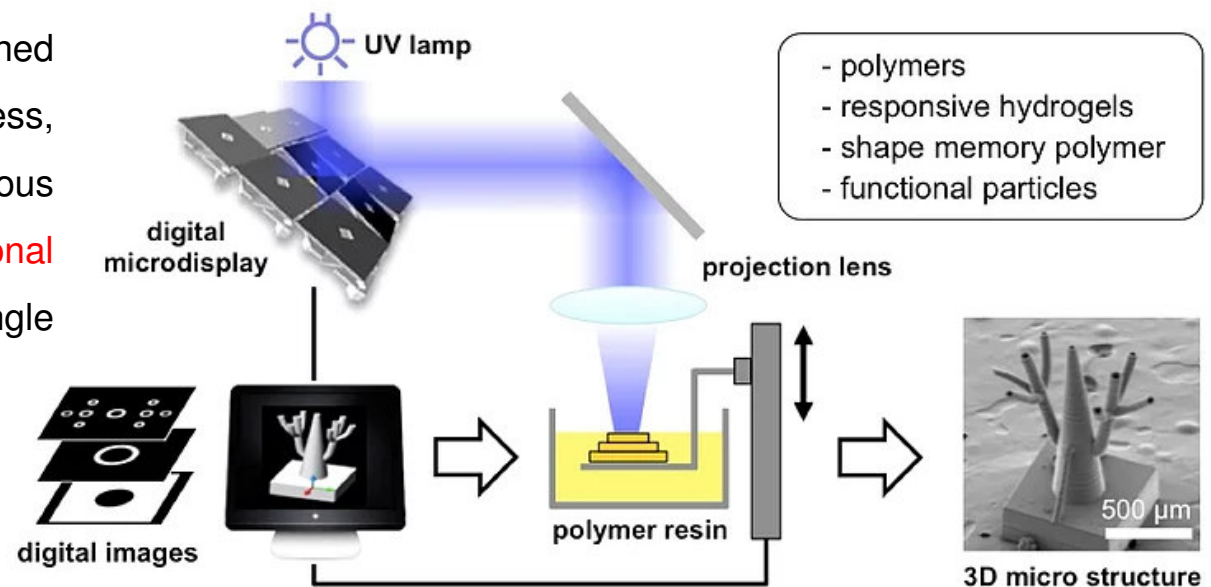
Rheological model

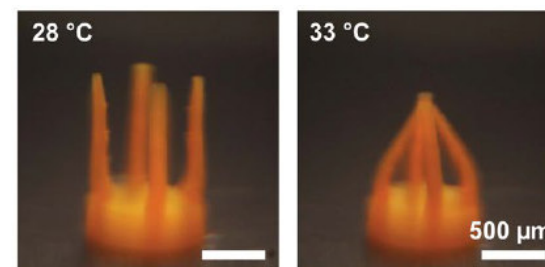
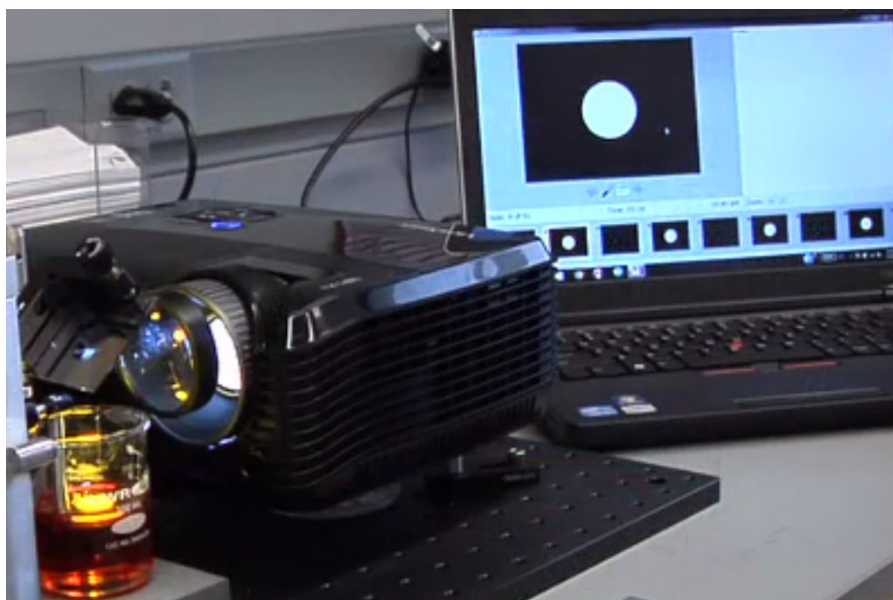
- Mechanistic kinetic model
- Degree of cure as a function of time and space
- Correlation with mechanics: constitutive model

Latest technology in photopolymerization

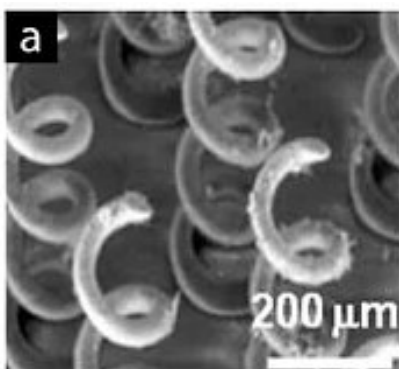
Projection Micro-StereoLithography (P μ SL)

- P μ SL utilizes the most advanced digital micro display technology as a dynamic mask generator.
- This technique combines advantages of conventional stereolithography (SLA) and **projection lithography**, allowing for rapid **photo-polymerization of the entire layer** with a flash of UV illumination at micro scale resolution.
- Light intensity is controlled at a single pixel level \rightarrow **crosslinking density** and thus material properties of the fabricated structure can be tailored with desired spatial distribution.
- Materials can be easily switched during the fabrication process, which enables heterogeneous integration of **multiple functional material** elements in a single process

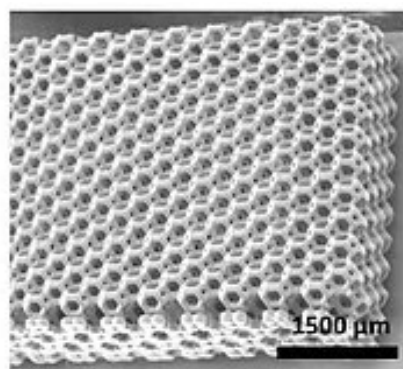




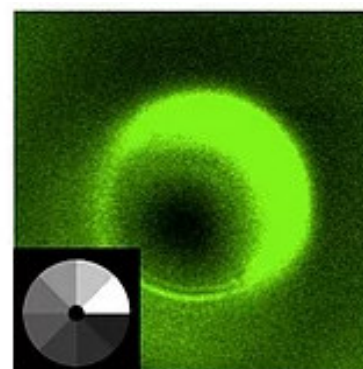
Responsive hydrogels



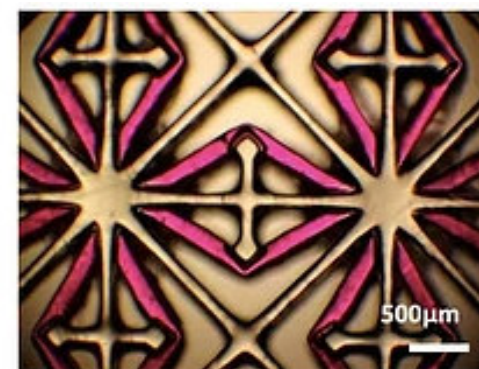
3D micro-structures



Scalable (μm to cm)



Functionally graded



Multi-material

- D. Han et al., Multimaterial Printing for Cephalopod-Inspired Light-Responsive Artificial Chromatophores, *ACS Appl. Mater. Interfaces* 2021, Jan. 3, 2021. <https://doi.org/10.1021/acsami.0c17623>

6. CONCLUSIONS

- ❑ Polymers are particularly **suitable to be Additively Manufactured**
- ❑ Several **technologies** are available
 - powder bed fusion
 - photopolymerization
 - material extrusion
 - material jetting, binder jetting
 - sheet lamination
- ❑ The size of the printed object can range **from macro to micro**
- ❑ Many different polymers can be printed; also **multimaterial** AM is possible
- ❑ The printing process can be often **precisely controlled**
- ❑ **New AM technologies** are emerging, allowing to obtain unprecedented materials

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